

Phospholipids used for in vitro biopharmaceutical characterization of oral drugs

PRC Webinar

14th of January 2026

Ann-Christin Jacobsen, PhD



In vitro biopharmaceutical characterization

- **Biopharmaceutics:** Predicting if and to what extent a drug is absorbed

In vitro methods

Dissolution testing

Permeation studies

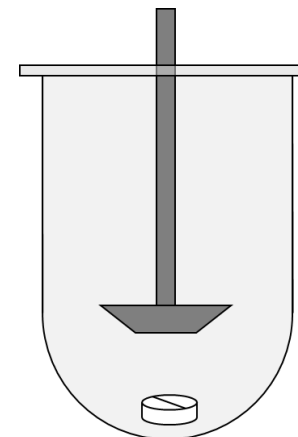
Phospholipids???

Combined dissolution/permeation studies

Phospholipids in dissolution testing

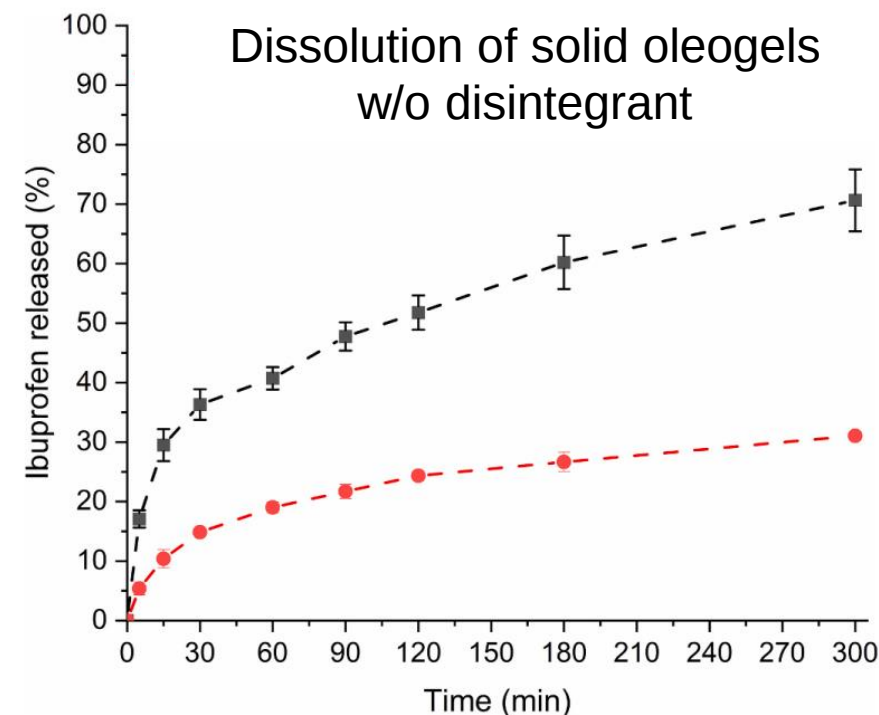
Dissolution testing

- Following the dissolution of a drug (from a dosage form) over time in aqueous medium
- Various instruments (e.g. USP paddle/basket)
- **Parameters to consider:**
 - Dissolution volume
 - Stirring speed / flow
 - Geometry
 - Sampling technique
 - Medium
- **Medium:**
 - pH
 - Buffer type, salt(s)
 - Enzymes, bile salts, **PLs**



Nernst-Brunner equation

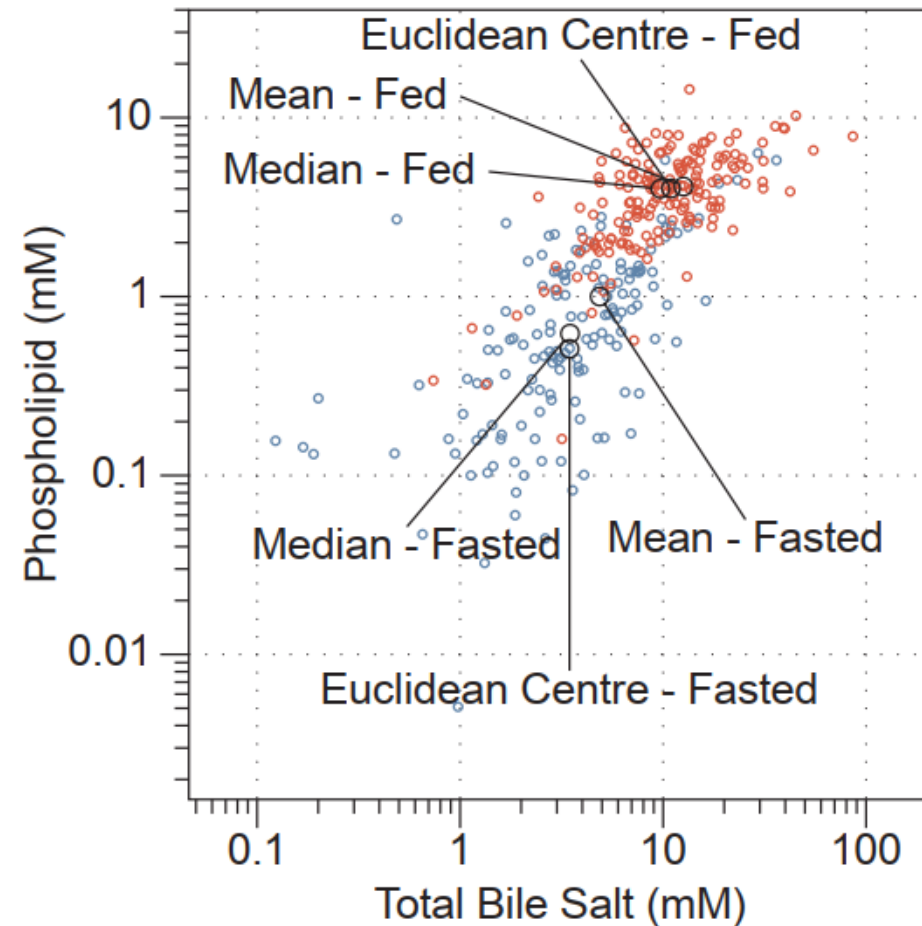
$$\frac{dC}{dt} = \frac{D \times A}{V} \times \frac{c_s - c_t}{h}$$



Phospholipids in intestinal fluids

- Phospholipids are present in (gastro)intestinal fluids
- Sources of PLs in GI fluids:
 - Bile
 - Food
- PLs in Bile:
 - >95 % PC
 - ~ 4:1 ratio between bile salts and PLs
- Assemblies:
 - Mixed bile salt-phospholipid micelles
 - Vesicles (fed)
- PL concentration in GI-fluids:
 - Intra- and interindividual variations
 - Variations with prandial state

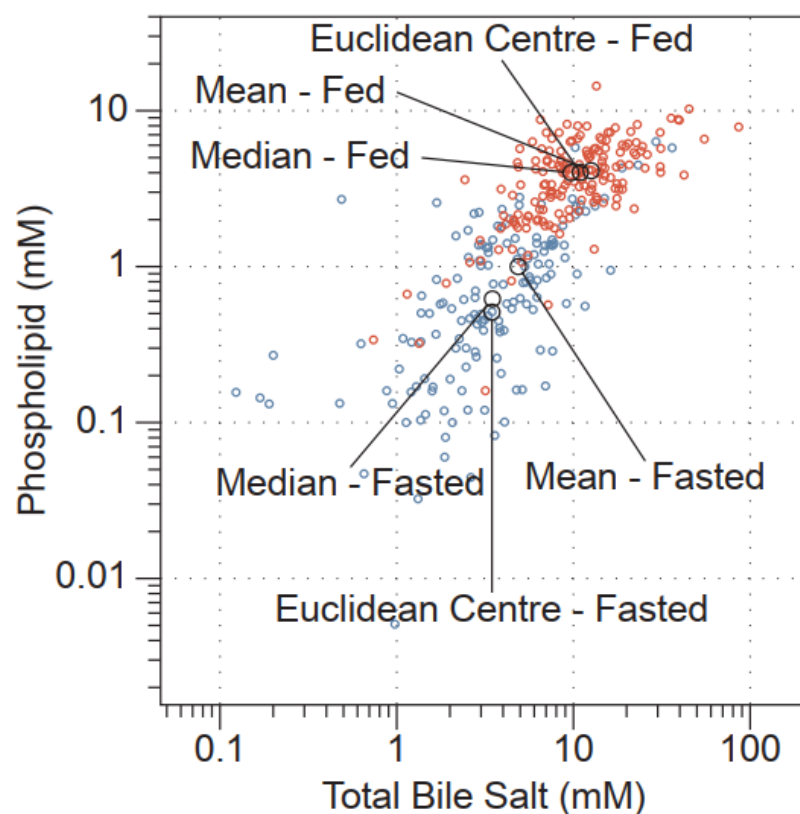
Solubilization of poorly soluble drugs!



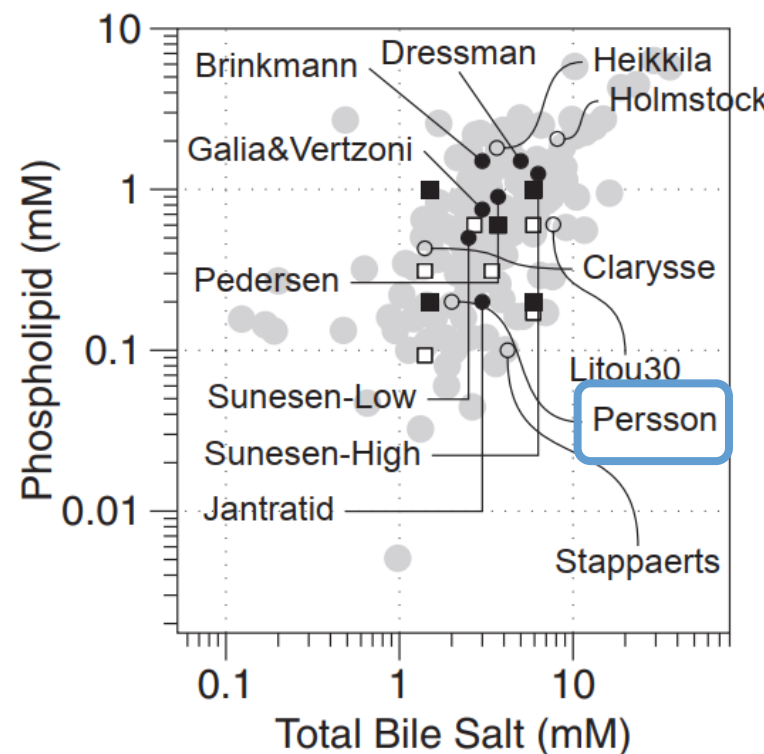
Phospholipids in intestinal fluids

Compared to literature

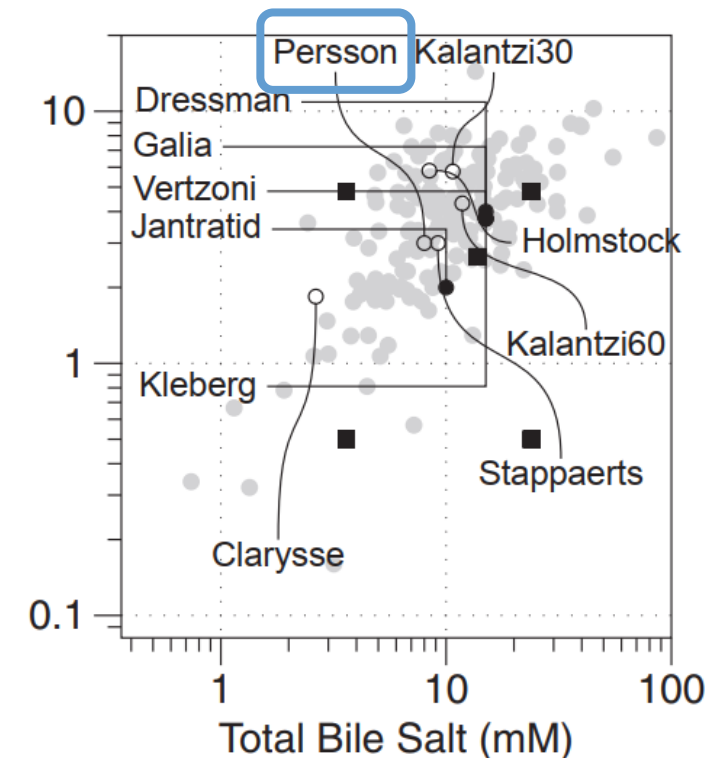
Riethorst et al. 2016 data



Fasted-state HIF



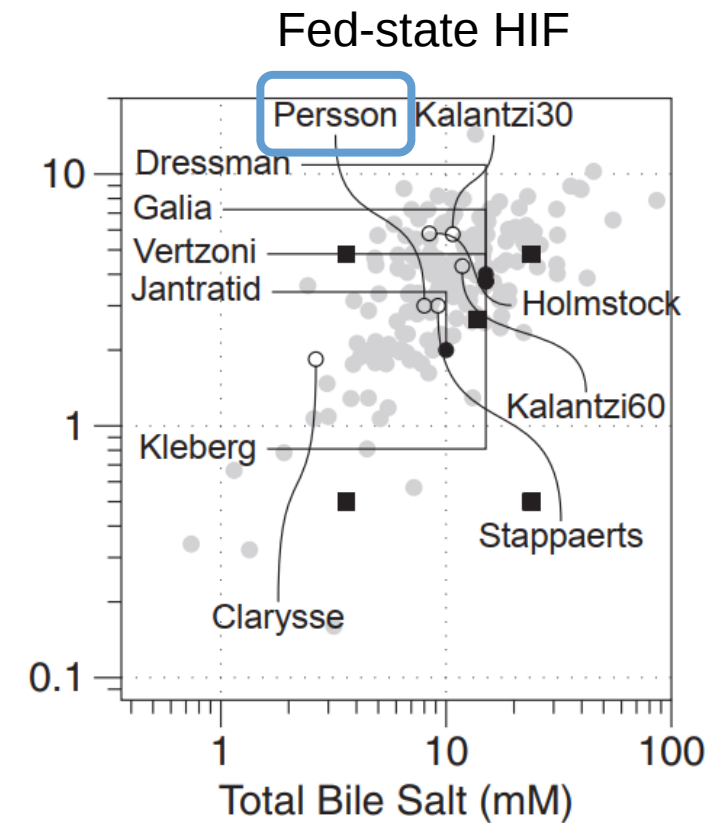
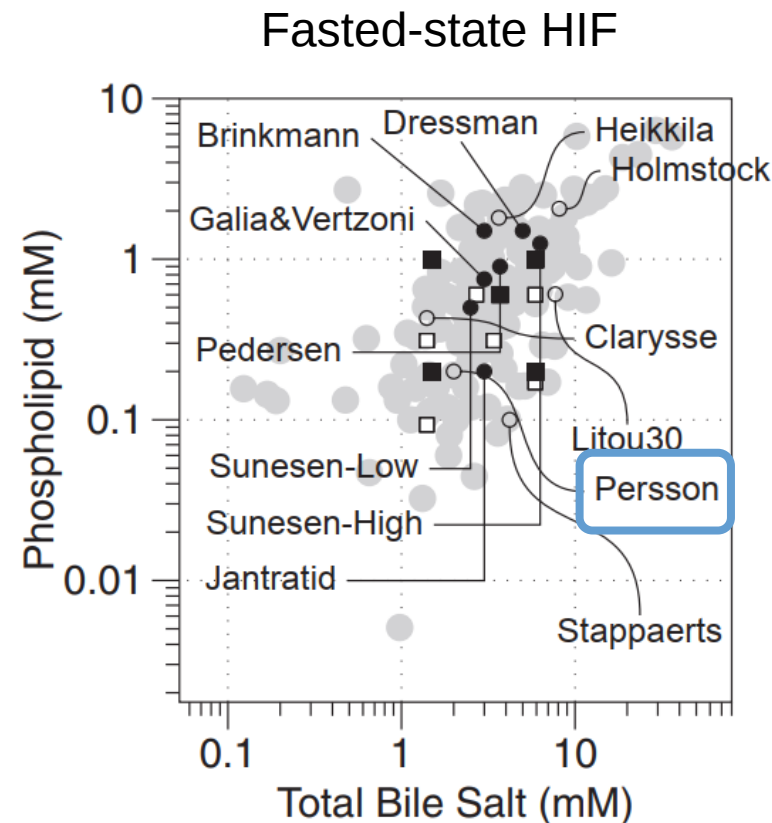
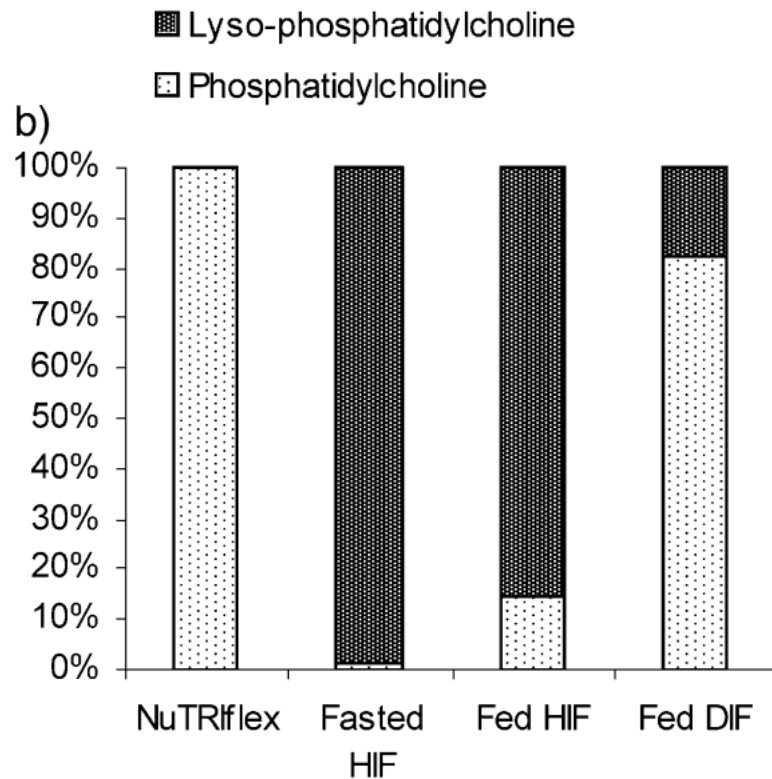
Fed-state HIF



Quantification of PLs (in most studies):
Enzymatic assay (choline)

Phospholipids in intestinal fluids

Compared to literature



Simulated intestinal fluids

- Mimicking (human) intestinal fluids
- Lecithin and bile salts (sodium taurocholate; NaTC)
- First described in 1998 by Galia et al. DOI: 10.1023/a:1011910801212
- Few variants

	FaSSIF V1*	FeSSIF V1*	FaSSIF V2*	FeSSIF V2*	FaSSIF V3**
Cholesterol (mM)	-	-	-	-	0.2
Glycerol monooleate	-	-	-	5	-
Lecithin (mM)	0.75	3.75	0.15	2	0.035
Lysolecithin (mM)	-	-	-	-	0.315
Sodium glycocholate (mM)	-	-	-	-	1.4
Sodium oleate (mM)	-	-	-	0.8	0.315
Sodium taurocholate (mM)	3	15	3	10	1.4
pH	6.5	5.0	6.5	5.8	6.7

* Commercially available from biorelevant.com, ** composition from Klumpp et al. 2020, DOI: 10.1016/j.ejps.2019.105138

Solubilization of poorly soluble drugs – simulated intestinal fluids

Table 2. Apparent Solubility and Intrinsic Dissolution Rate^a

compound	apparent solubility ($\mu\text{g/mL}$)				IDR ($\mu\text{g/min/cm}^2$)			
	pH 6.5		pH 5.0		pH 6.5		pH 5.0	
	FaSSIF _{blk}	FaSSIF	FeSSIF _{blk}	FeSSIF	FaSSIF _{blk}	FaSSIF	FeSSIF _{blk}	FeSSIF
albendazole	0.85 ± 0.36	1.9 ± 0.0	1.1 ± 0.0	6.1 ± 0.1	0.09 ± 0.01	0.13 ± 0.01	0.12 ± 0.01	0.30 ± 0.01
astemizole	19.2 ± 0.2	97.9 ± 2.1	248 ± 6	182 ± 1	1.80 ± 0.10	7.76 ± 0.17	22.7 ± 0.6	16.9 ± 0.1
carvedilol	46.0 ± 1.0	55.9 ± 1.0	243 ± 2	305 ± 2	4.40 ± 0.10	5.40 ± 0.09	23.2 ± 0.2	26.0 ± 0.2
cinnarizine	1.4 ± 2.5	13.4 ± 0.9	11.8 ± 0.2	112 ± 2	0.14 ± 0.25	0.33 ± 0.02	1.18 ± 0.17	4.44 ± 0.07
danazol	0.6 ± 0.0	8.4 ± 0.7	0.9 ± 0.1	28.8 ± 0.4	0.06 ± 0.01	0.16 ± 0.01	0.09 ± 0.01	0.95 ± 0.01
felodipine	1.2 ± 0.0	54.5 ± 3.7	1.1 ± 0.0	237 ± 1	0.12 ± 0.01	0.64 ± 0.04	0.11 ± 0.00	7.16 ± 0.01
glibenclamide	4.5 ± 1.1	4.7 ± 0.1	0.3 ± 0.0	2.5 ± 0.2	0.48 ± 0.11	0.41 ± 0.01	0.03 ± 0.00	0.10 ± 0.01
indomethacin	219 ± 78	443 ± 10	14.9 ± 0.2	109 ± 7	21.7 ± 7.8	33.5 ± 0.7	1.49 ± 0.02	4.63 ± 0.28
tamoxifen	5.9 ± 0.2	156 ± 2	22.6 ± 0.9	236 ± 13	0.58 ± 0.02	2.25 ± 0.03	2.21 ± 0.09	23.1 ± 1.3
tolfenamic acid	27.4 ± 0.5	63.0 ± 2.7	0.8 ± 0.0	41.0 ± 0.5	2.98 ± 0.06	4.01 ± 0.17	0.09 ± 0.01	1.31 ± 0.02
min	0.6	1.9	0.3	2.5	0.06	0.13	0.03	0.10
max	219	443	248	305	21.7	33.5	23.2	26.0

^a Apparent solubility and intrinsic dissolution rate (IDR) in fasted state simulated intestinal fluid (FaSSIF), fed state simulated fluid (FeSSIF) and their corresponding blank buffers without taurocholic acid and lecithin (FaSSIF_{blk} and FeSSIF_{blk}).

Solubilization of poorly soluble drugs – simulated intestinal fluids

Table 2. Apparent Solubility and Intrinsic Dissolution Rate^a

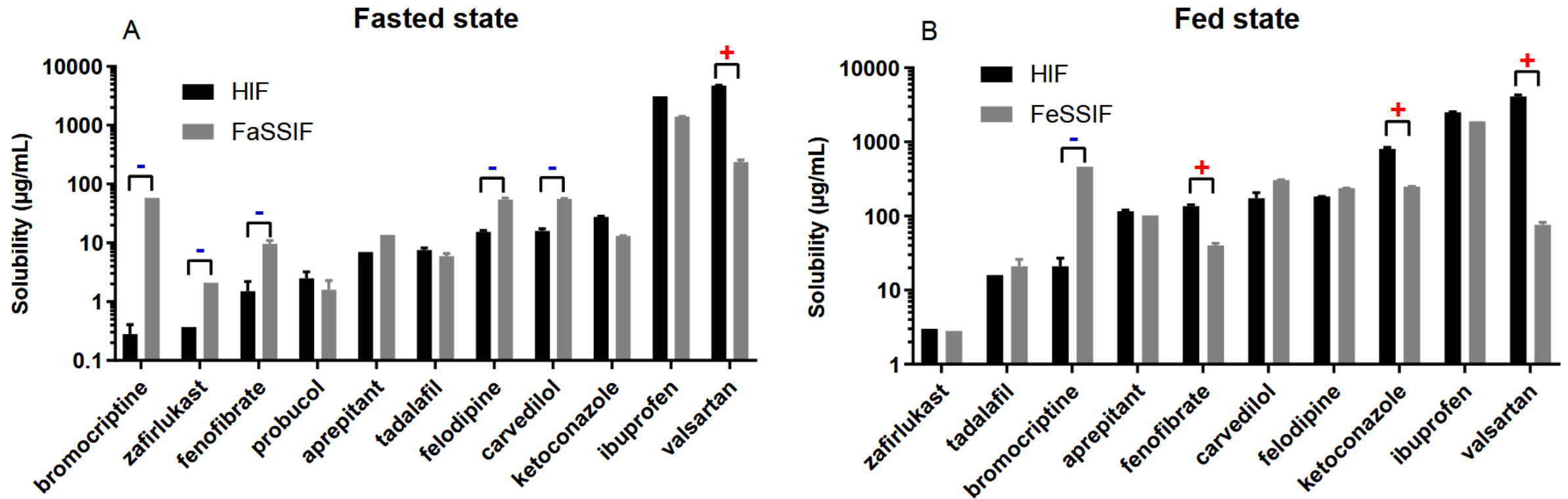
compound	apparent solubility ($\mu\text{g/mL}$)						IDR ($\mu\text{g/min/cm}^2$)			
	pH 6.5			pH 5.0			pH 6.5		pH 5.0	
	FaSSIF _{blk}	FaSSIF		FeSSIF _{blk}	FeSSIF		FaSSIF _{blk}	FaSSIF	FeSSIF _{blk}	FeSSIF
albendazole	0.85 ± 0.36	1.9 ± 0.0	2.24	1.1 ± 0.0	6.1 ± 0.1	5.55	0.09 ± 0.01	0.13 ± 0.01	0.12 ± 0.01	0.30 ± 0.01
astemizole	19.2 ± 0.2	97.9 ± 2.1	5.10	248 ± 6	182 ± 1	0.73	1.80 ± 0.10	7.76 ± 0.17	22.7 ± 0.6	16.9 ± 0.1
carvedilol	46.0 ± 1.0	55.9 ± 1.0	1.22	243 ± 2	305 ± 2	1.26	4.40 ± 0.10	5.40 ± 0.09	23.2 ± 0.2	26.0 ± 0.2
cinnarizine	1.4 ± 2.5	13.4 ± 0.9	9.57	11.8 ± 0.2	112 ± 2	9.49	0.14 ± 0.25	0.33 ± 0.02	1.18 ± 0.17	4.44 ± 0.07
danazol	0.6 ± 0.0	8.4 ± 0.7	14.00	0.9 ± 0.1	28.8 ± 0.4	32.00	0.06 ± 0.01	0.16 ± 0.01	0.09 ± 0.01	0.95 ± 0.01
felodipine	1.2 ± 0.0	54.5 ± 3.7	45.42	1.1 ± 0.0	237 ± 1	215.5	0.12 ± 0.01	0.64 ± 0.04	0.11 ± 0.00	7.16 ± 0.01
glibenclamide	4.5 ± 1.1	4.7 ± 0.1	1.04	0.3 ± 0.0	2.5 ± 0.2	8.33	0.48 ± 0.11	0.41 ± 0.01	0.03 ± 0.00	0.10 ± 0.01
indomethacin	219 ± 78	443 ± 10	2.02	14.9 ± 0.2	109 ± 7	7.32	21.7 ± 7.8	33.5 ± 0.7	1.49 ± 0.02	4.63 ± 0.28
tamoxifen	5.9 ± 0.2	156 ± 2	26.44	22.6 ± 0.9	236 ± 13	10.44	0.58 ± 0.02	2.25 ± 0.03	2.21 ± 0.09	23.1 ± 1.3
tolfenamic acid	27.4 ± 0.5	63.0 ± 2.7	2.30	0.8 ± 0.0	41.0 ± 0.5	51.25	2.98 ± 0.06	4.01 ± 0.17	0.09 ± 0.01	1.31 ± 0.02
min	0.6	1.9	1.04	0.3	2.5	0.73	0.06	0.13	0.03	0.10
max	219	443	45.42	248	305	215.5	21.7	33.5	23.2	26.0

^a Apparent solubility and intrinsic dissolution (FeSSIF) and their corresponding blank buffers

ate (IDR) in fasted state
out taurocholic acid and I

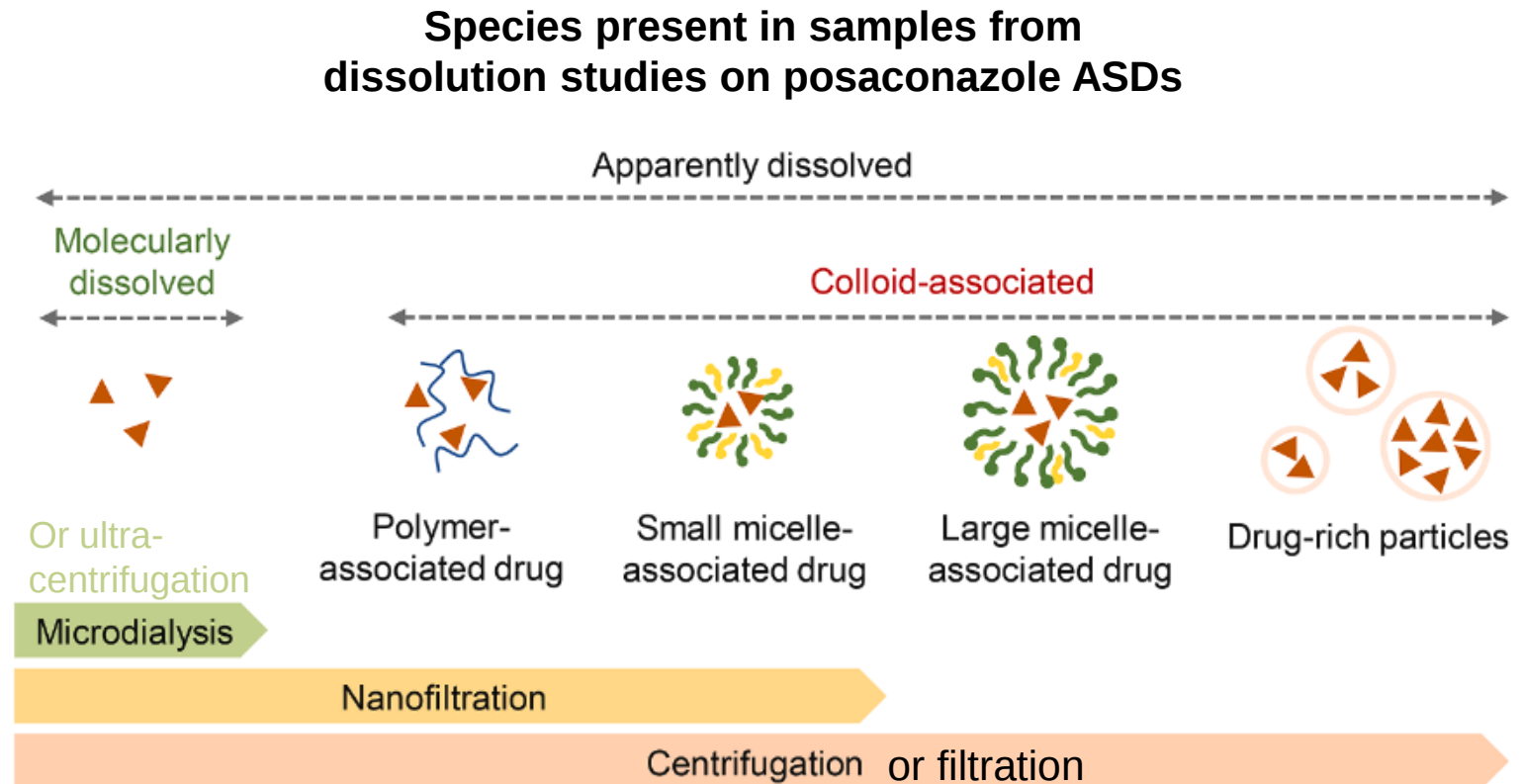
imulated intestinal fluid (FaSSIF), fed state simulated fluid
ithin (FaSSIF_{blk} and FeSSIF_{blk}).

Simulated vs human intestinal fluids



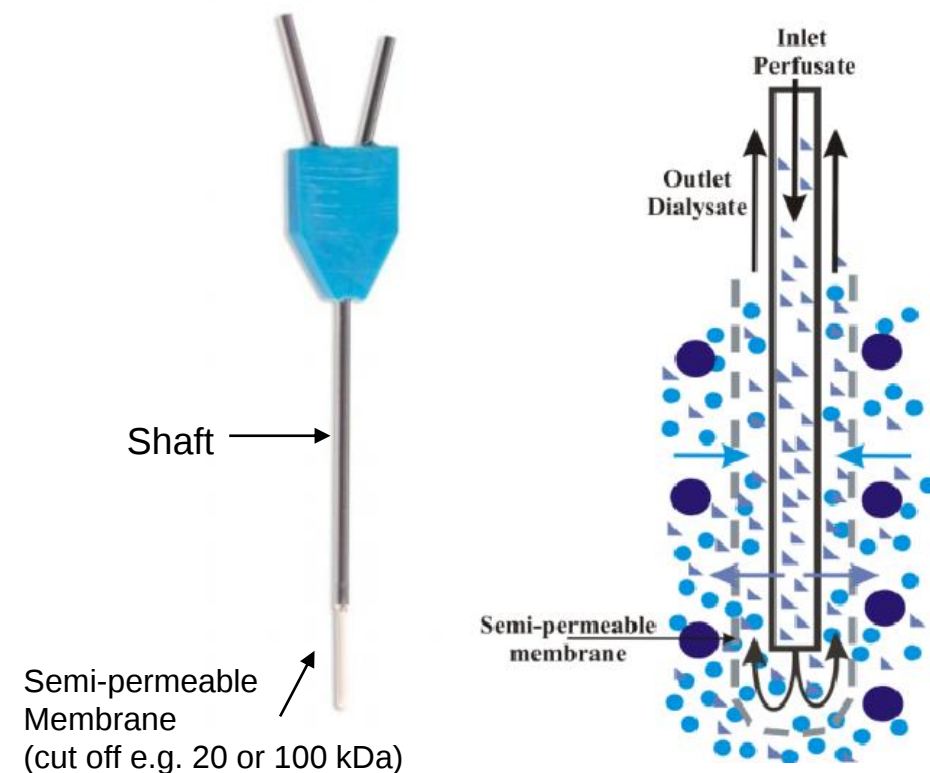
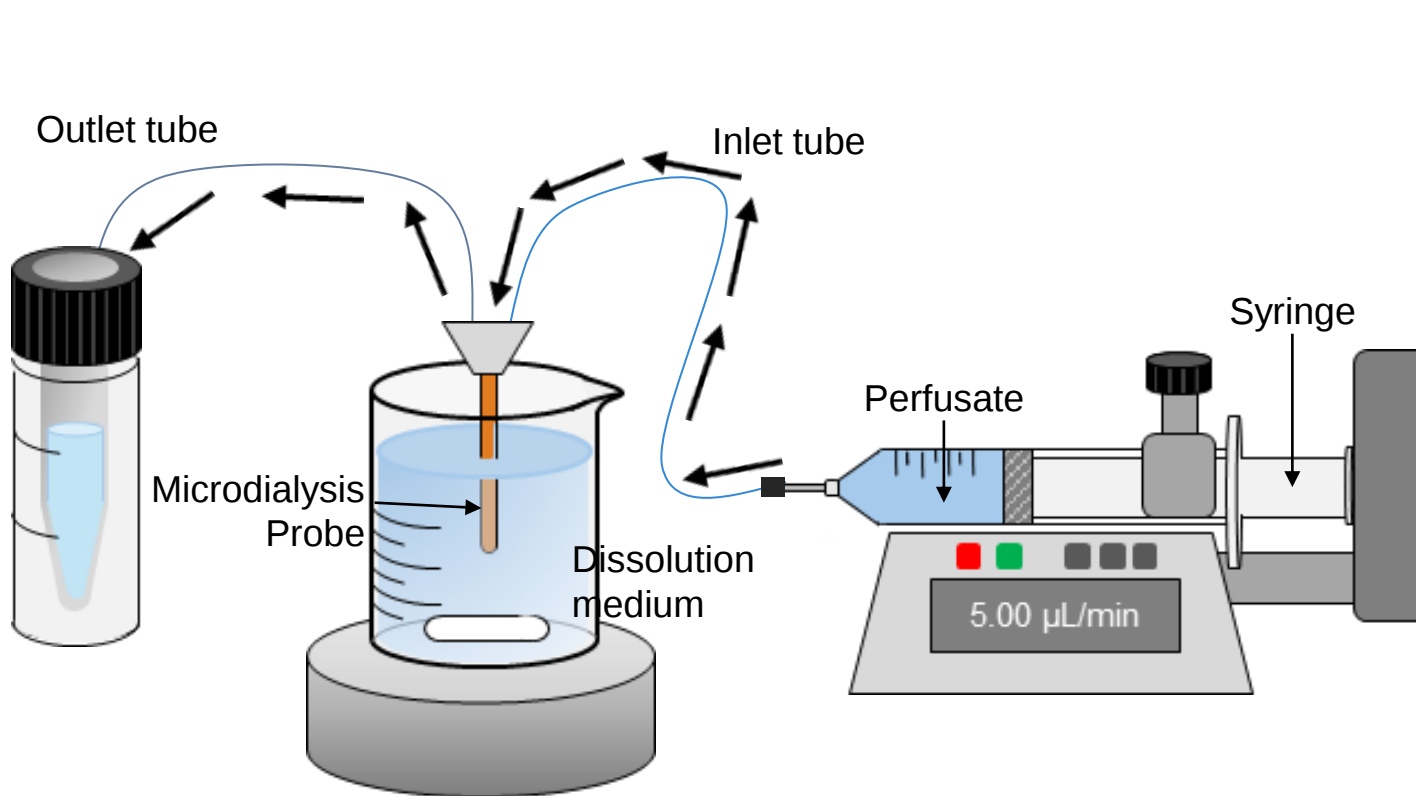
Increased solubility in presence of mixed micelles: what are we measuring?

- Typical sample preparation during solubility or dissolution studies:
 - Filtration or bench-top centrifugation
 - Dilution
- Alternative sampling / sample preparation:
 - Microdialysis
 - Ultra-centrifugation



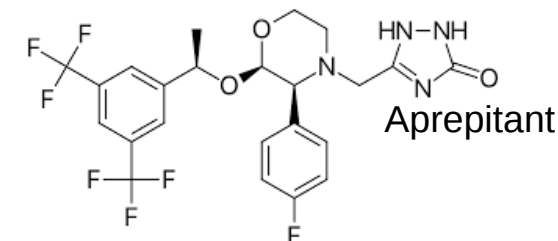
Microdialysis sampling

- Commonly used for minimal invasive tissue sampling (e.g., drug concentration in brain)
- Can separate molecules (or very small assemblies) from colloids (e.g. mixed micelles)



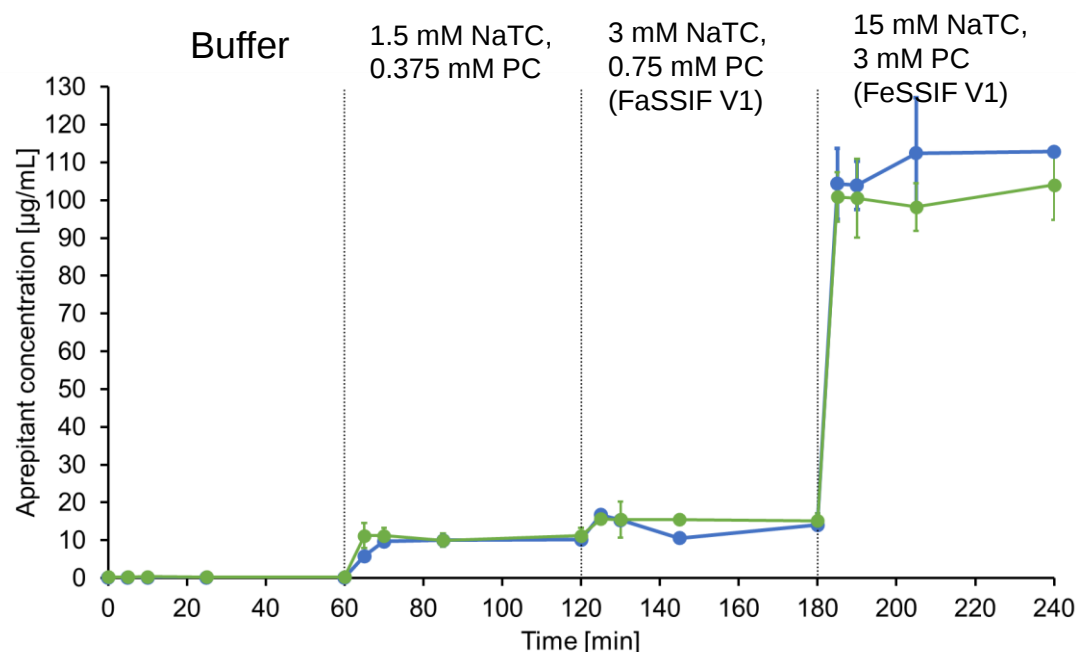
Example: Apparent vs molecular solubility of aprepitant in simulated intestinal fluids

- Very poorly soluble in aqueous media (e.g., $\sim 0.25 \mu\text{g/ml}$ in PBS pH 6.5)
- Simulated intestinal fluids substantially increase apparent solubility
- Microdialysis sampling:
 - Significant increase in molecular solubility in FeSSIF V1 (pH 6.5)

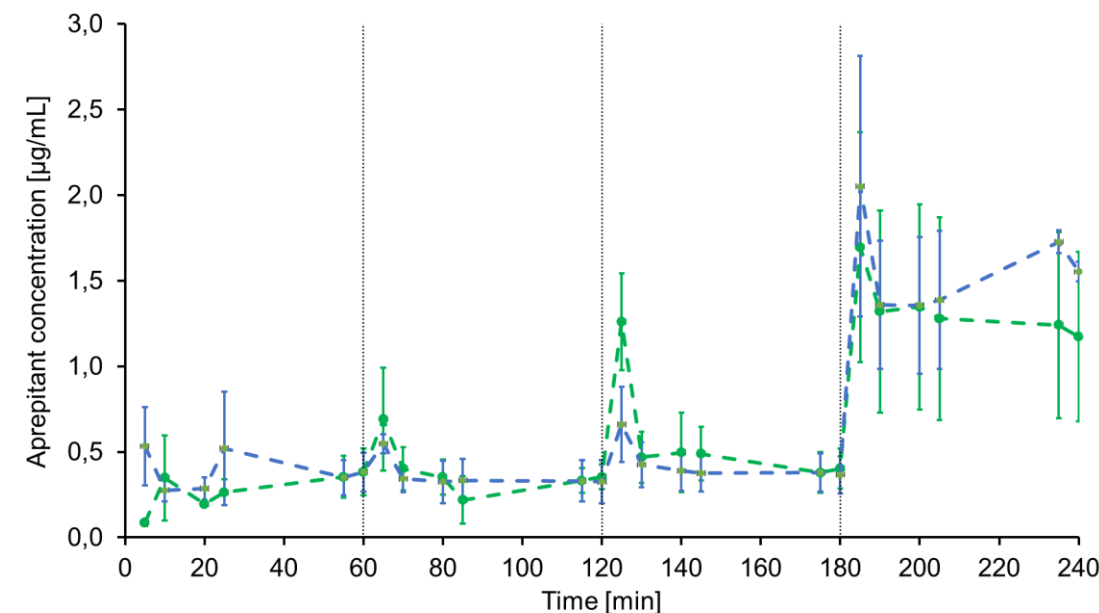


20 kDa cut-off
100 kDa cut off

Bench-top centrifugation

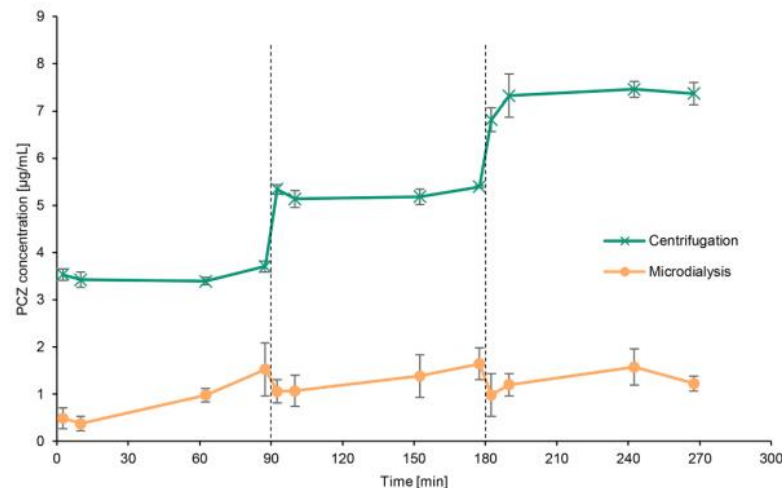


Microdialysis sampling



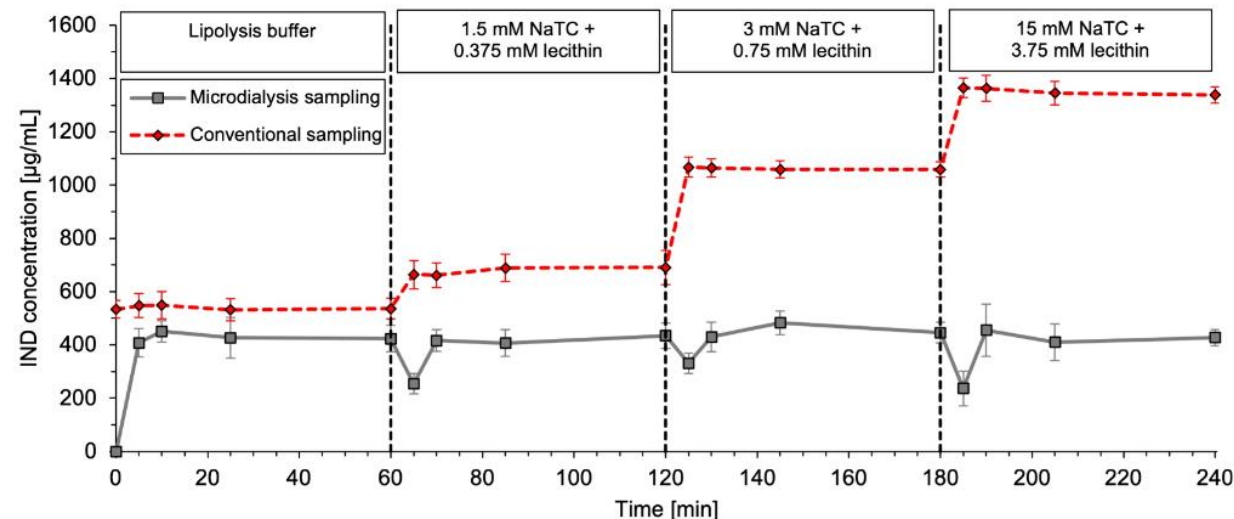
Comparison – Molecular solubility in presence of (mixed) micelles

Posaconazole

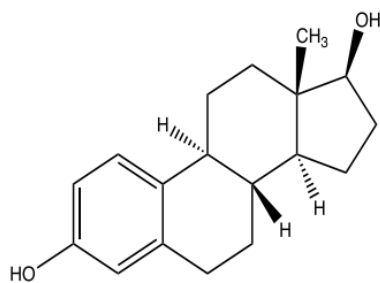


Holzem et al. 2022, *Eur. J. Pharm. Sci.*,
DOI:10.1016/j.ejps.2022.106166

Indomethacin



Tønning et al. 2025, *Mol. Pharm.*, DOI: 10.1021/acs.molpharmaceut.5c00640



(a) 17β-Estradiol
(E2)

Ultra-centrifugation

Without PS80: Crystalline and amorphous solubility of E2
2.2 and 13.9 µg/mL, respectively.

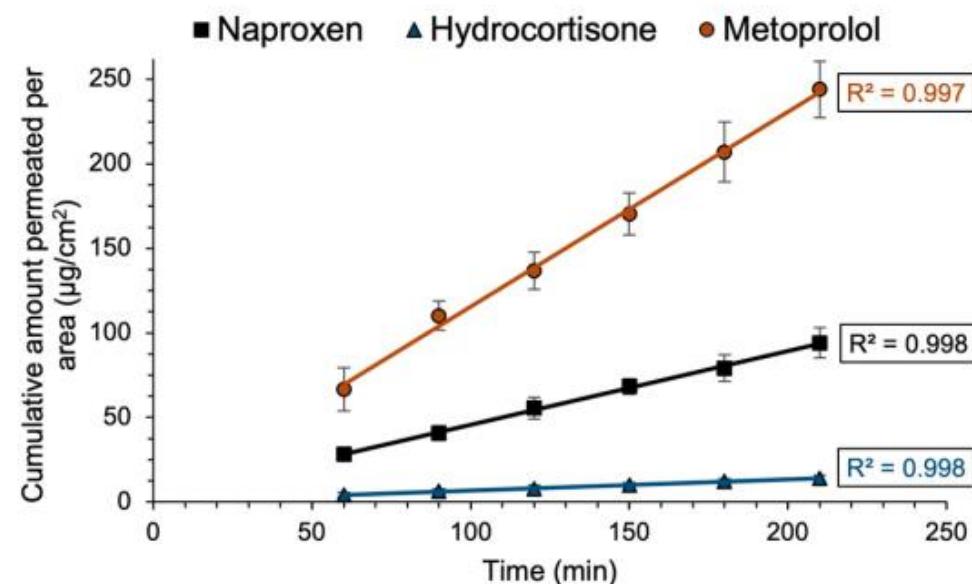
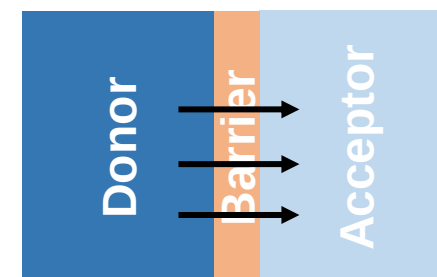
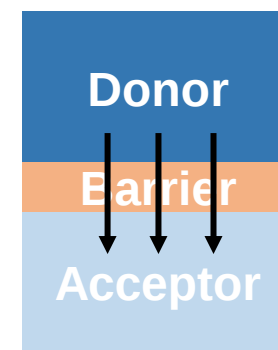
With 1% PS80: Crystalline and amorphous solubility of E2
4.6 µg/mL and 15.7 µg/mL

Arce et al. 2020, *Mol. Pharm.*, DOI:10.1021/acs.molpharmaceut.0c00599

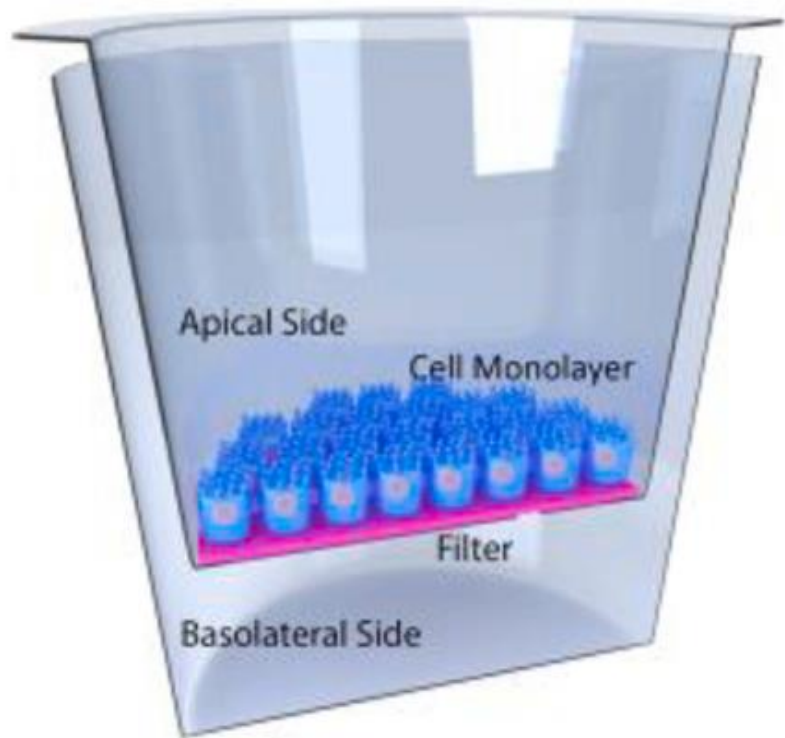
Phospholipids in permeation studies

Permeation studies

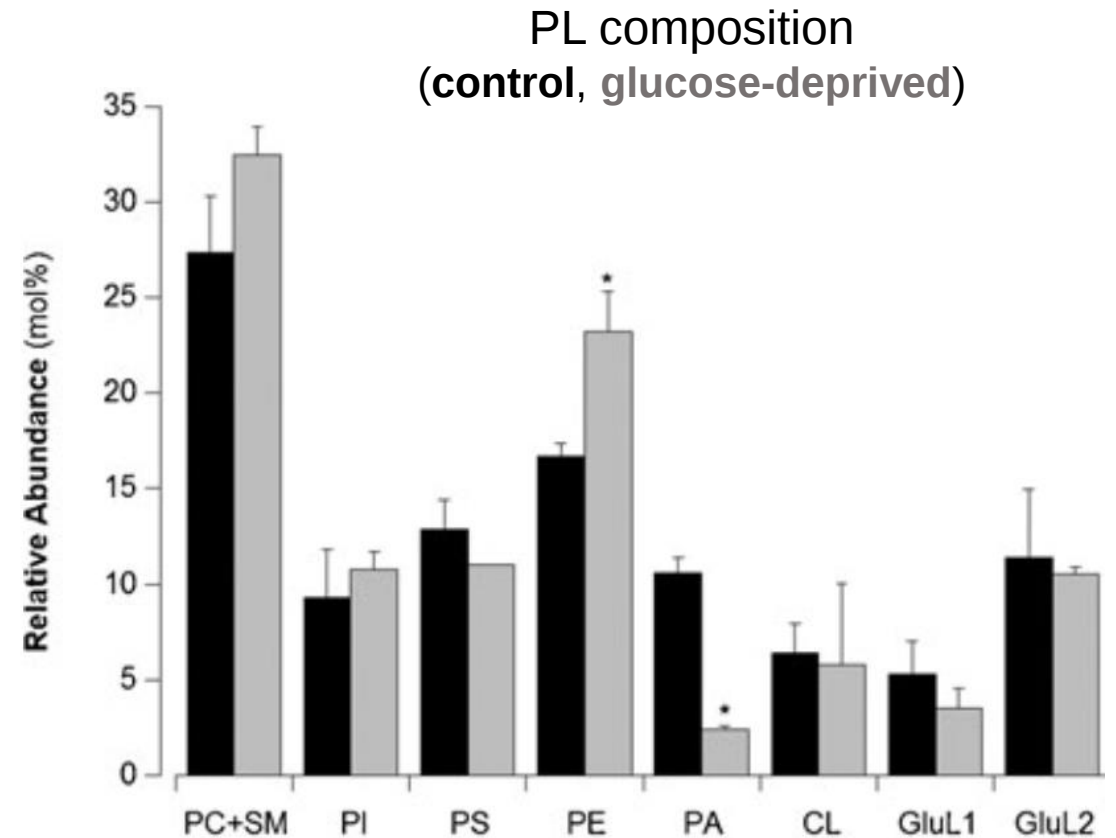
- Following the movement of drug molecules from one compartment (donor) across a barrier to another compartment (acceptor/receiver)
- **Barriers:**
 - tissues (e.g., intestinal tissue from rat),
 - cells (e.g. **Caco-2** monolayers),
 - artificial barriers w/o PLs



Caco-2 permeability assay



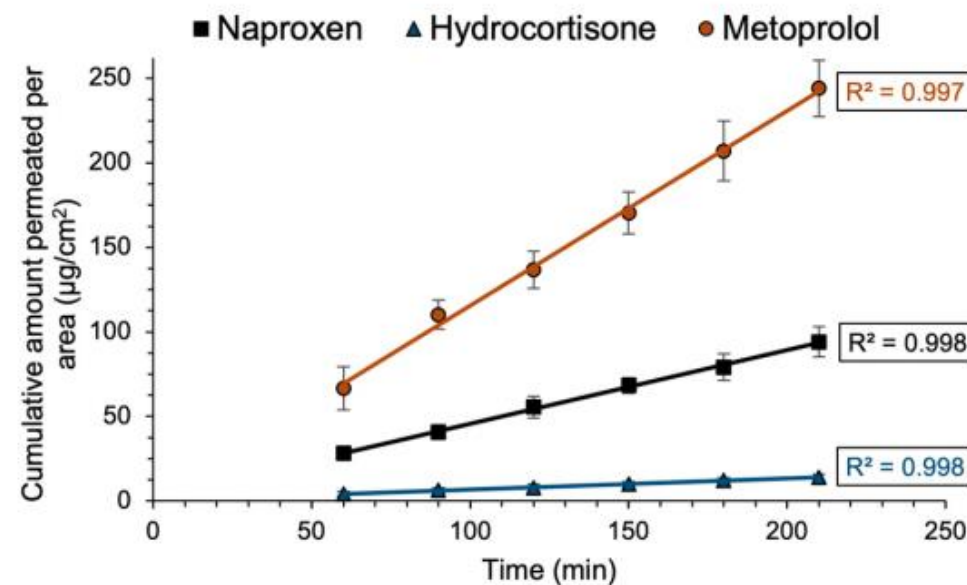
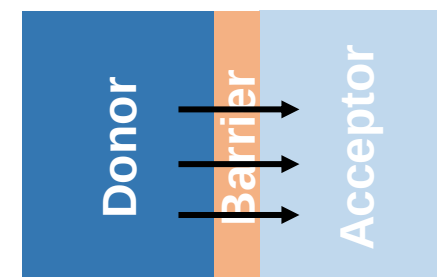
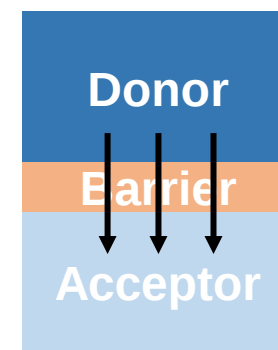
Jarc et al. 2019, *J. Pharm. Pharmacol.*,
DOI: 10.1111/jphp.13111



Monteiro-Cardoso et al. 2014, *J. Bioenerg. Biomembr.*,
DOI: 10.1007/s10863-013-9531-y

Permeation studies

- Following the movement of drug molecules from one compartment (donor) across a barrier to another compartment (acceptor/receiver)
- **Barriers:**
 - tissues (e.g., intestinal tissue from rat),
 - cells (e.g., Caco-2 monolayers),
 - artificial barriers w/o PLs
- **Artificial barriers with PLs:**
 - PAMPA
 - PVPA
 - Permeapad®

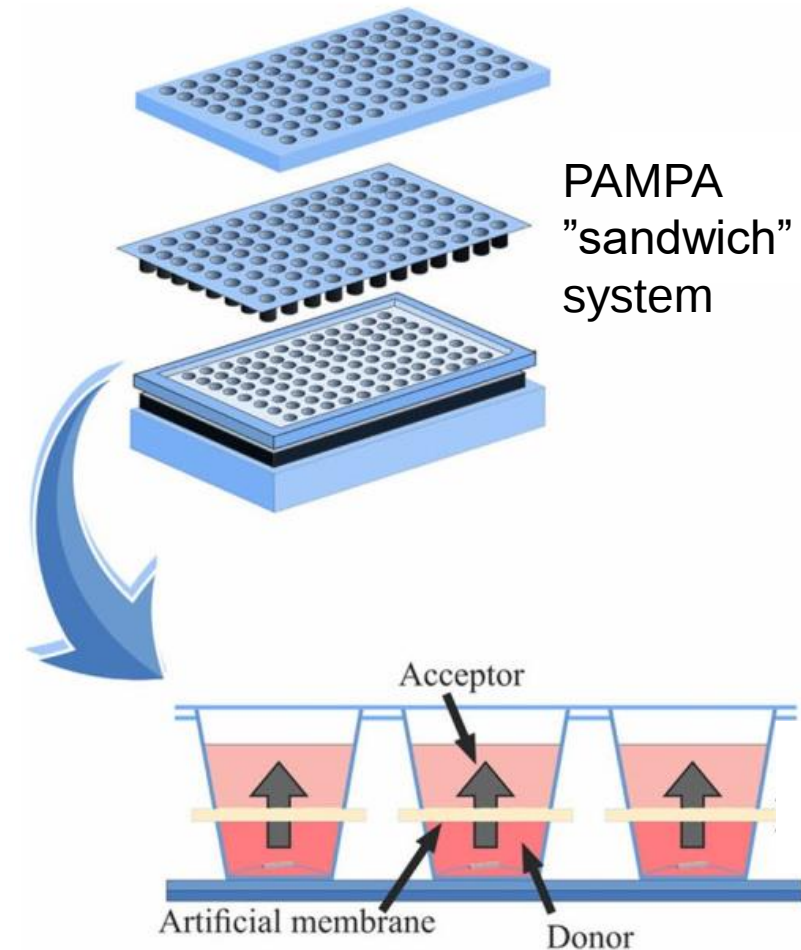


PAMPA

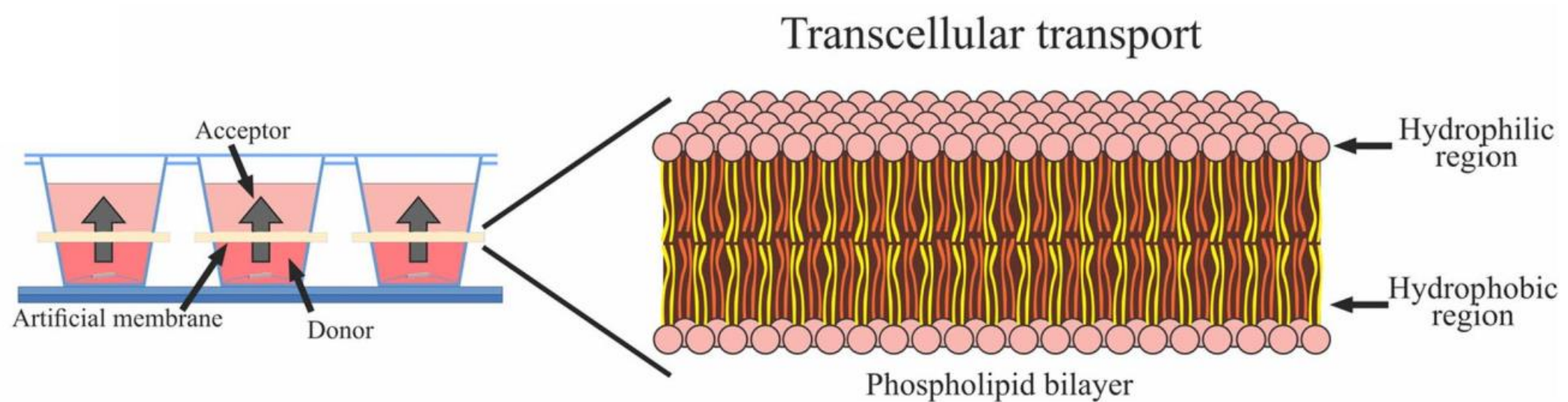
- Developed by Kansy and co-workers (F. Hoffmann-La Roche) in 1998
- Barrier composition: A **filter** impregnated with **phospholipids** dissolved in an **organic solvent**
- 96-well format most common
- Several variants

Table 1
Characteristics of PAMPA assays.

Assay short name	Barrier target	Support filter	Solvent	Membrane composition	Lit. ref
Egg-PAMPA	GI-tract	Hydrophobic PVDF (125 µm)	n-Dodecane	10% egg lecithin	(Kansy et al., 1998)
HDM-PAMPA	GI-tract	Polycarbonate 10 µm, non-tissue culture treated	n-Hexadecane	n-Hexadecane	(Wohnsland and Faller, 2001)
BM-PAMPA	GI-tract	Hydrophobic PVDF (125 µm)	1.7-Octadiene	3% of a phospholipid mixture	(Sugano et al., 2001)
DOPC-PAMPA	GI-tract	Hydrophilic PVDF (125 µm)	n-Dodecane	1% egg lecithin	(Zhu et al., 2002)
PAMPA-DS	GI-tract	Hydrophobic PVDF (125 µm)	n-Dodecane	2% DOPC	(Avdeef et al., 2001)
BBB-PAMPA	BBB	Hydrophobic PVDF (125 µm)	n-Dodecane	20% (phospholipid mixture)	(Bermejo et al., 2004)
PAMPA-skin	Skin	Hydrophobic PVDF (125 µm)	n-Dodecane	2% polar brain lipids	(Di et al., 2003)
Precoated PAMPA	GI-tract	PVDF (0.45 µm)	70% silicon oil, 30% iso-propyl myristate	70% silicon oil, 30% iso-propyl myristate	(Ottaviani et al., 2006)
			Hexane	Lipid/oil/lipid trilayer	(Chen et al., 2008)

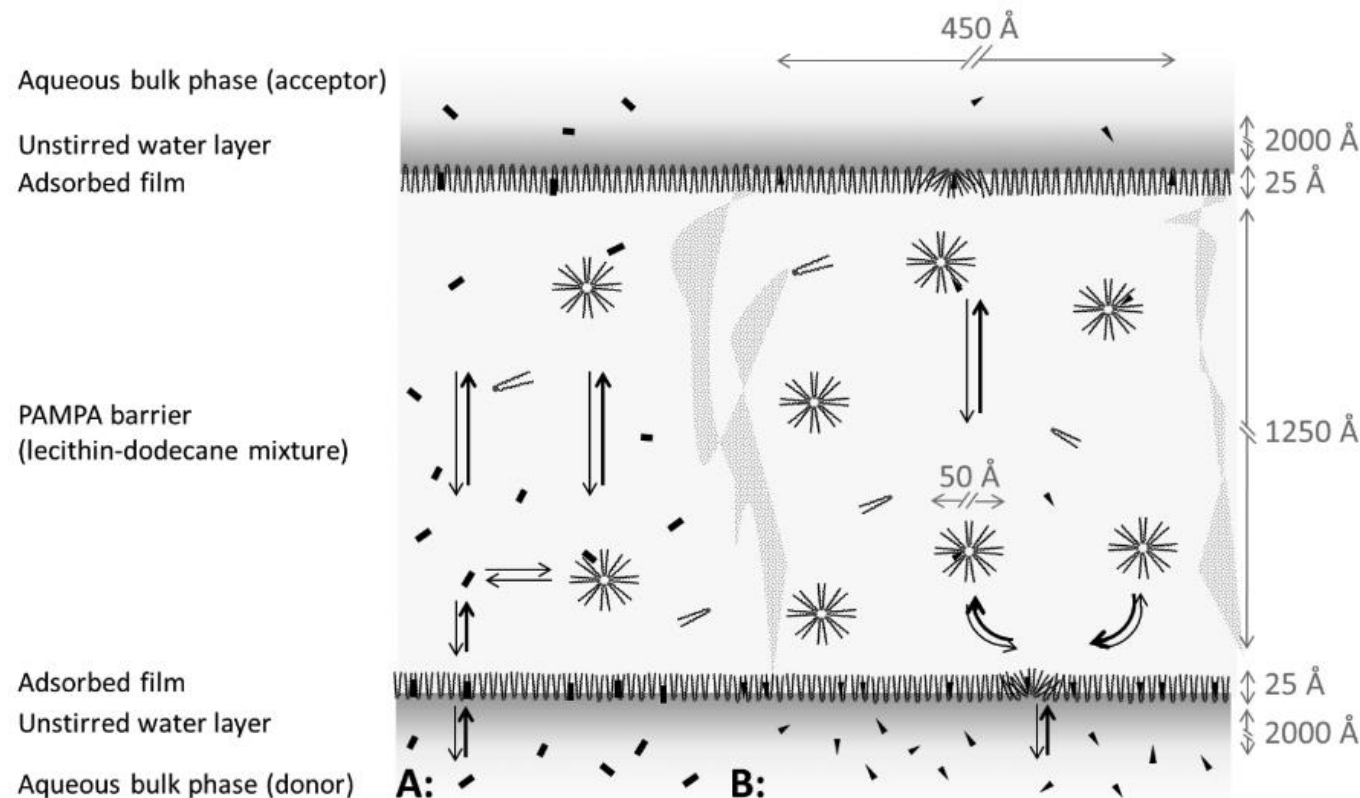


PAMPA barrier structure



Modified from Biondo et al. 2023, *Pharm. Res.* DOI: 10.1007/s11095-023-03499-9

PAMPA barrier structure

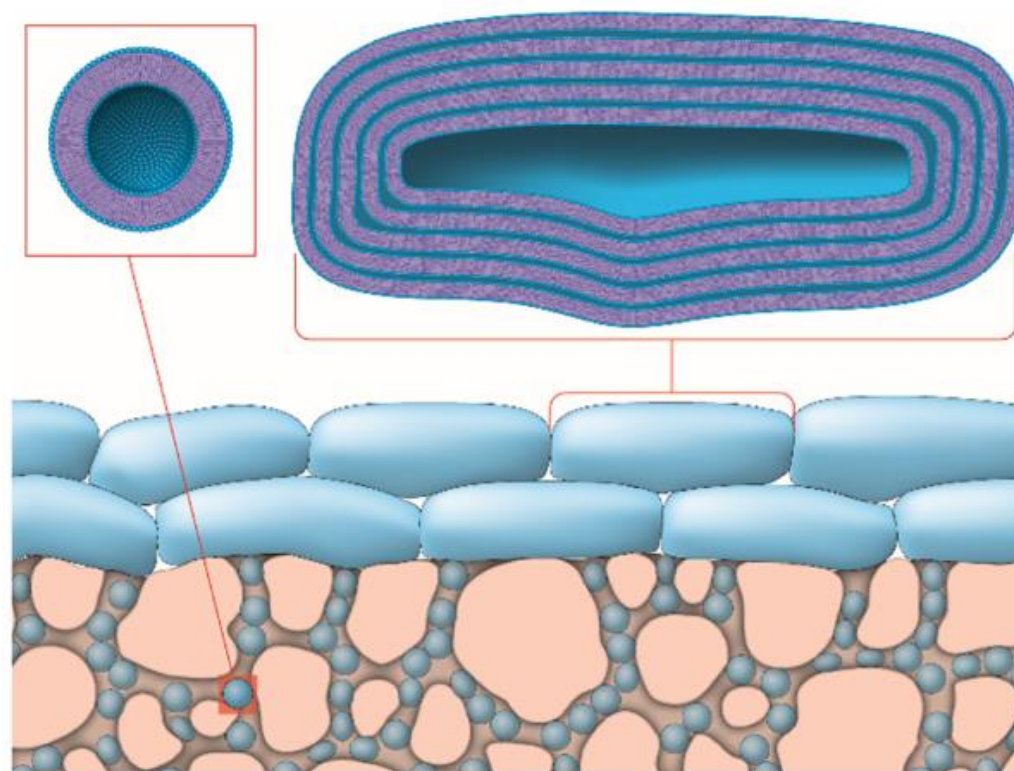


Reverse micelles: $d=6.9$ and 4.6 nm

- ^{31}P , ^1H NMR study
- Egg lecithin (>60% PC) in dodecane, PVDF filter
- Dodecane concentration:
 - low $\rightarrow$ inversion from a bilayer to inverted hexagonal phase.
 - high (>10 mol %) $\rightarrow$ formation of an isotropic phase
- PVDF filter: unfavorable for providing aperture for BLM (rough and hydrophilic).
- Reverse micelles of PC and PE in dodecane
 - Carriers for amphiphilic molecules (e.g., propranolol)
- Monolayer at PAMPA–lipid solution/water interface is plausible but could not be verified due to excess bulk (3000:1).

PVPA

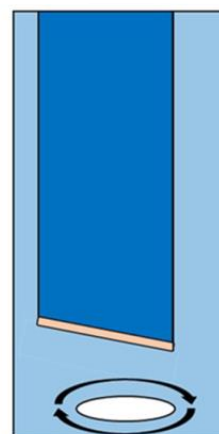
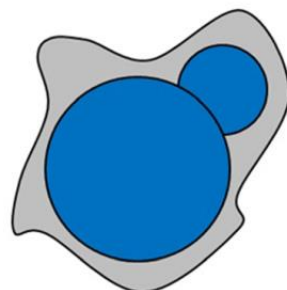
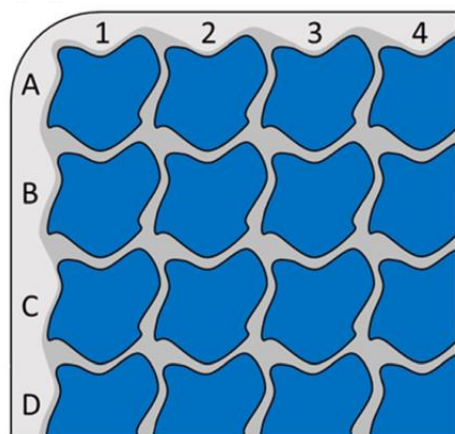
- Developed by Flaten, Brandl (University of Tromsø) and co-workers in 2006
- Barrier composition:
Liposomes of different sizes/lamellarity deposited on a filter support in a multi-step process including centrifugation
- Format: Transwell (6-, 12-well)
- Several variants



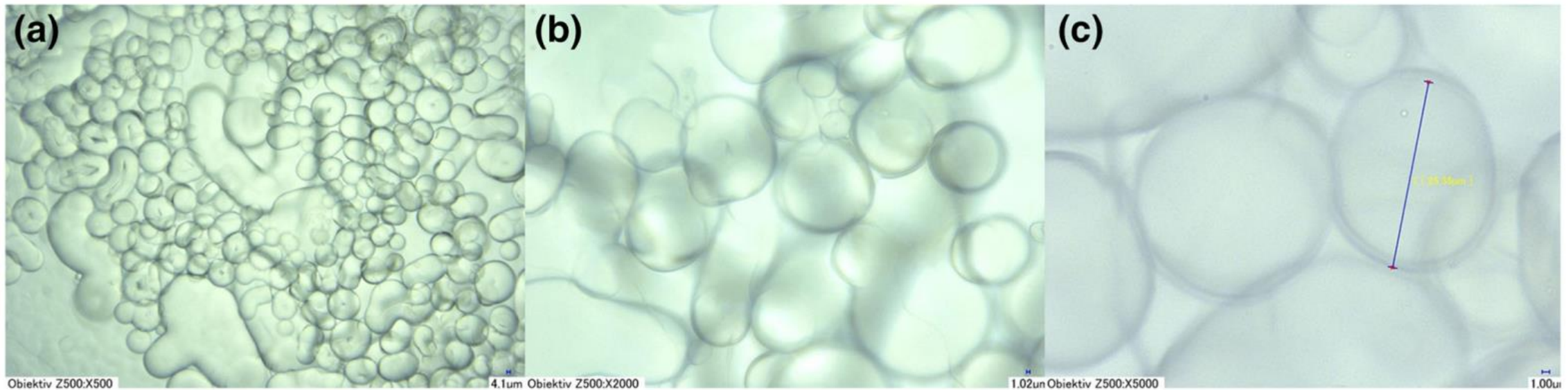
Berben et al. 2017, *Eur. J. Pharm. Sci.*,
DOI: 10.1007/s11095-023-03499-9

Permeapad®

- Developed by di Cagno and Bauer-Brandl (University of Southern Denmark) in 2015
- Barrier composition:
A sandwich of cellulose-hydrate membranes (support sheets) and **dry phospholipid**
- Formats:
 - Compatible with most permeation devices (side-by-side cells, Franz cells)
 - 96-well device

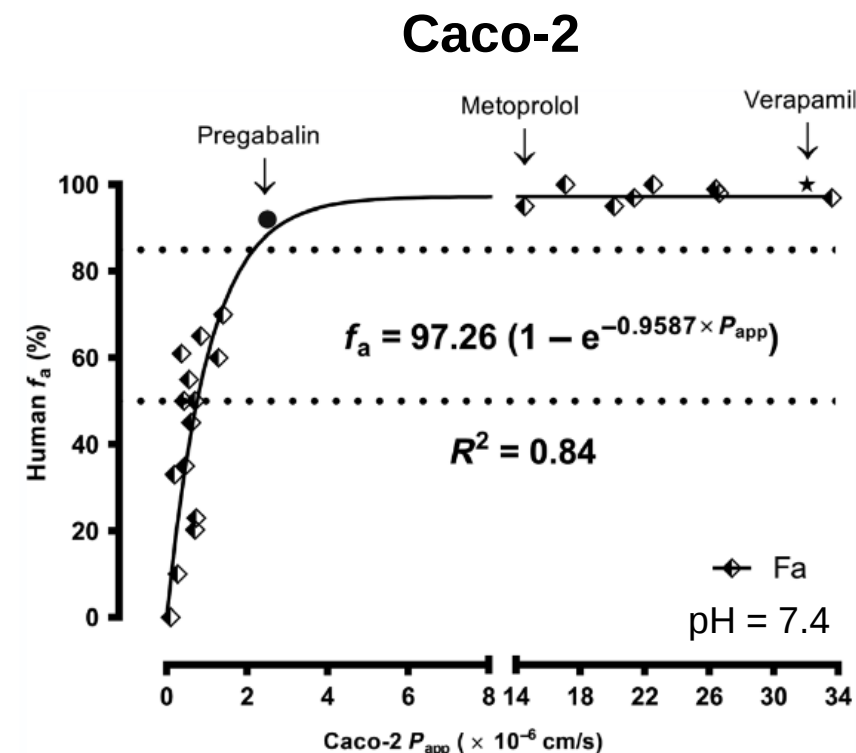


Permeapad® barrier structure

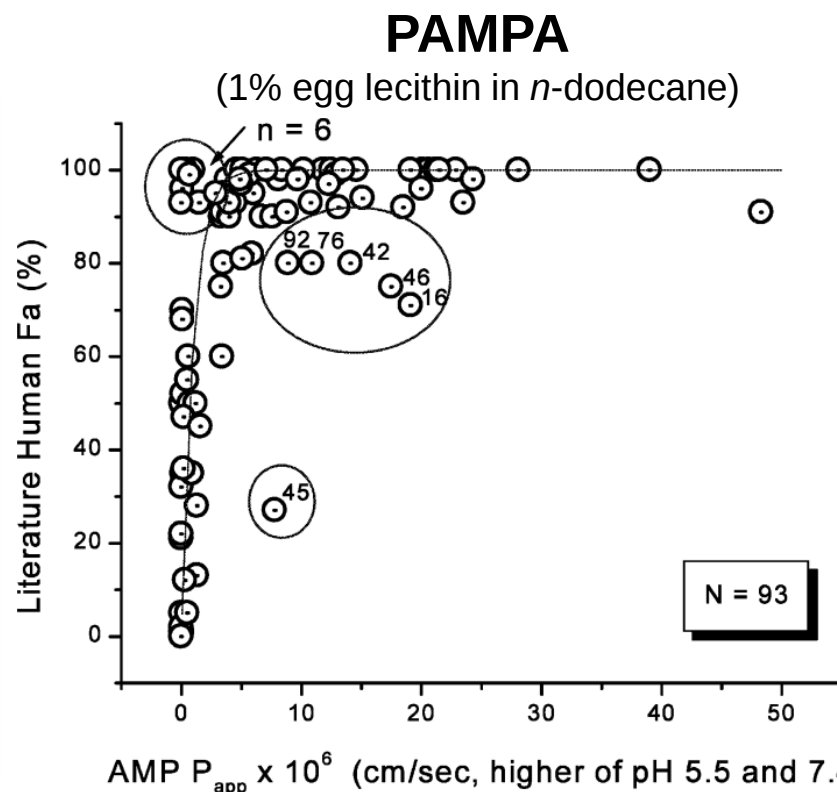


Oral absorption vs in vitro permeability

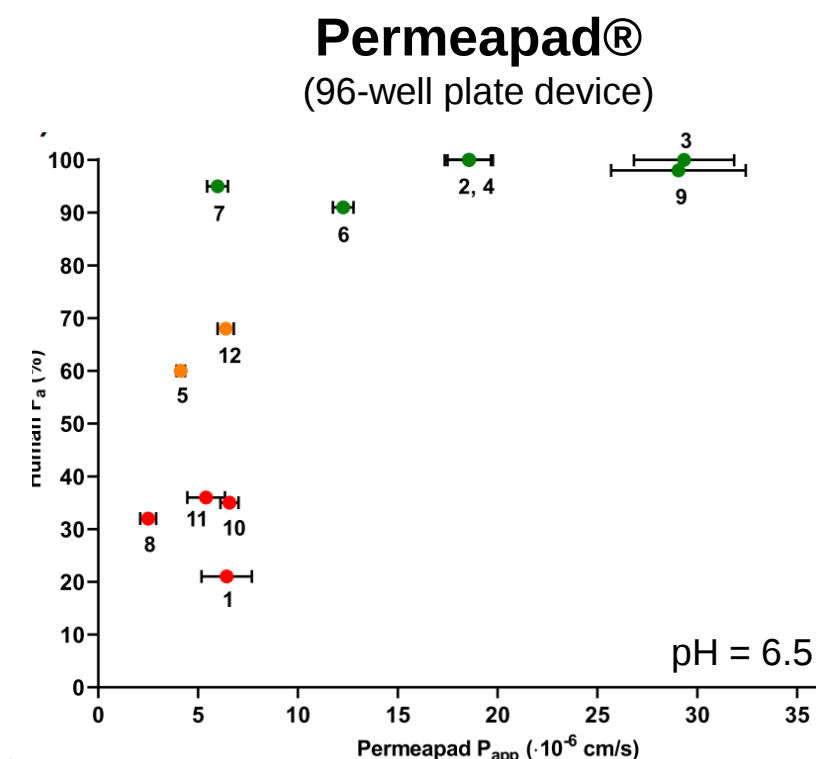
- Fraction absorbed in humans (f_a) vs. apparent permeability (P_{app} in $\times 10^{-6}$ cm/s)
- Steep slope region (low-moderate absorption), plateau region (high absorption)



Jarc et al. 2019, *J. Pharm. Pharmacol.*,
DOI: 10.1111/jphp.13111



Zhu et al. 2002, *Eur. J. Med. Chem.*
DOI: 10.1016/s0223-5234(02)01360-0



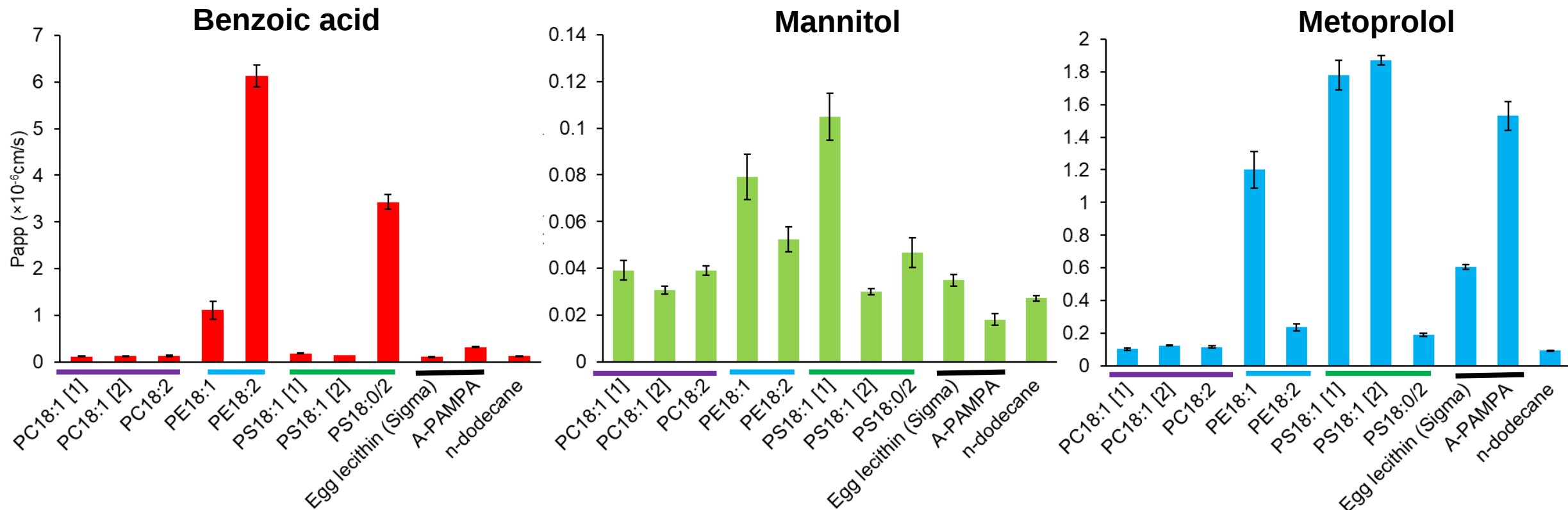
Jacobsen et al. 2020, *Pharm. Res.*,
DOI: 0.1007/s11095-020-02807-x

Influence of phospholipid composition on in vitro permeability determined with PAMPA, PVPA and Permeapad®

PAMPA: Influence of phospholipid composition

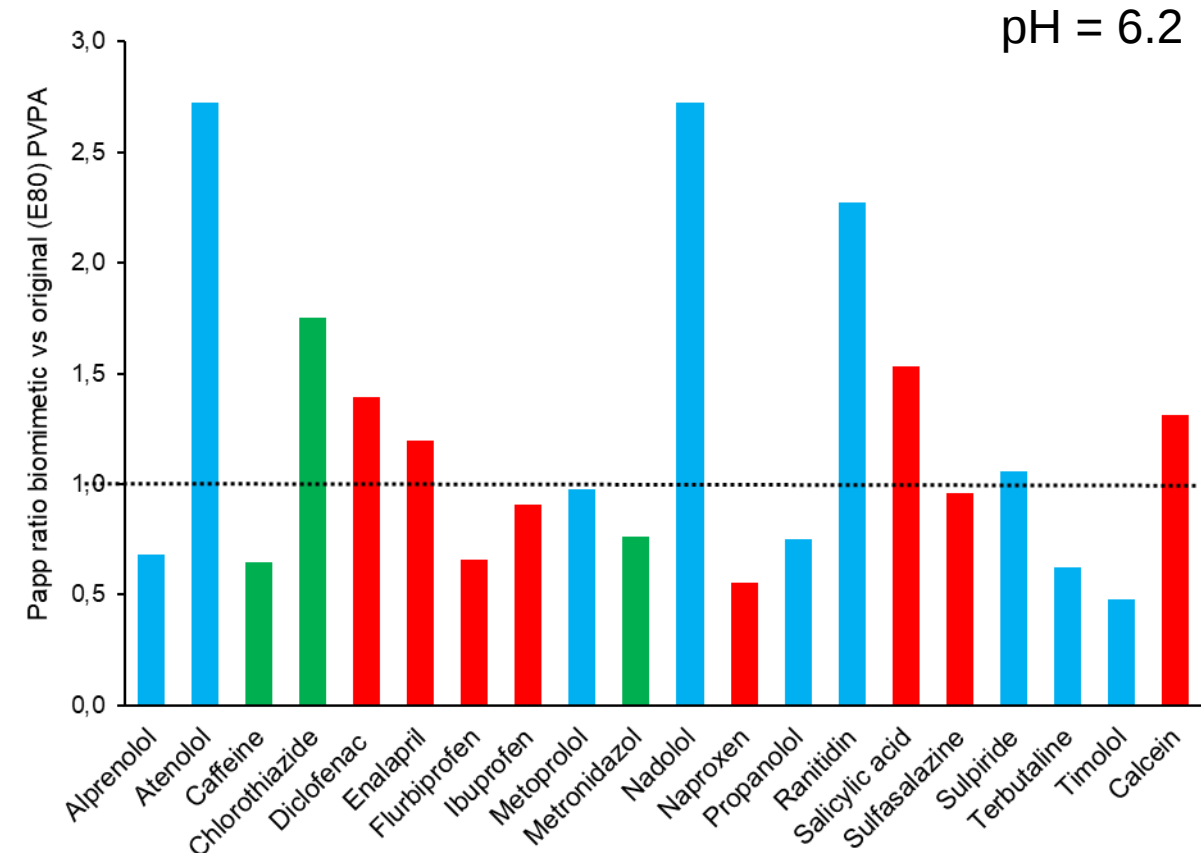
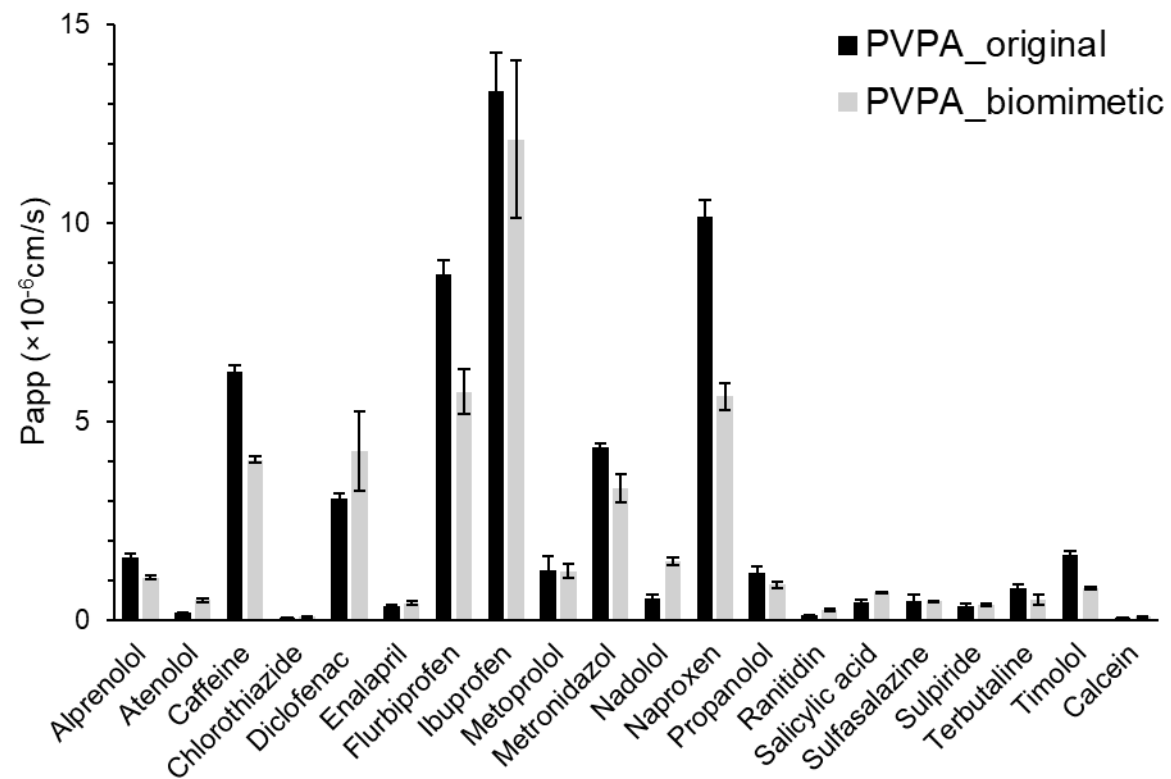
- 5 % phospholipid (w/v) in *n*-dodecane
- Either pure **PC**, **PE**, **PS** (varied unsaturation)
- Or mixture: **Egg lecithin**, **A-PAMPA** (52 % PC18:1, 18 % PS18:1, 30 % cholesterol)

pH = 6.5



PVPA: Influence of phospholipid composition

- **Original PVPA:** 100% phosphatidylcholine from egg (Lipoid E80)
- **Biomimetic PVPA:** 26.5% phosphatidylcholine, 26.5% phosphatidylethanolamine, 7% phosphatidylserine, 7% phosphatidylinositol and 33% cholesterol

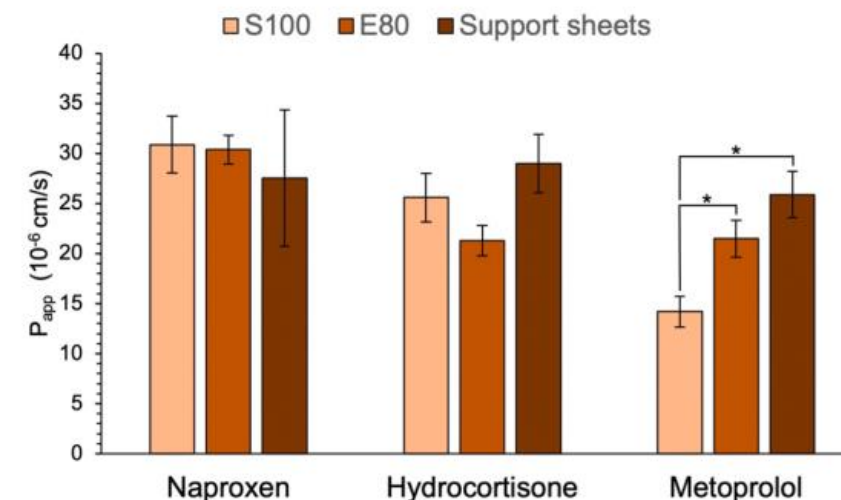
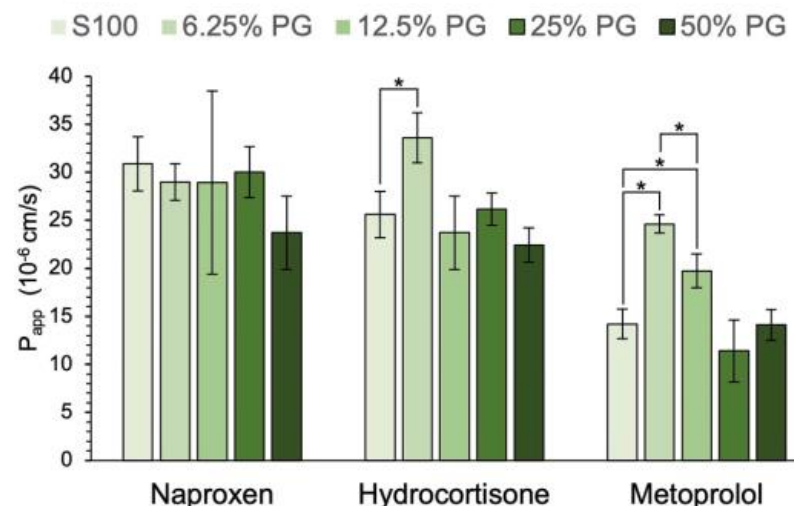
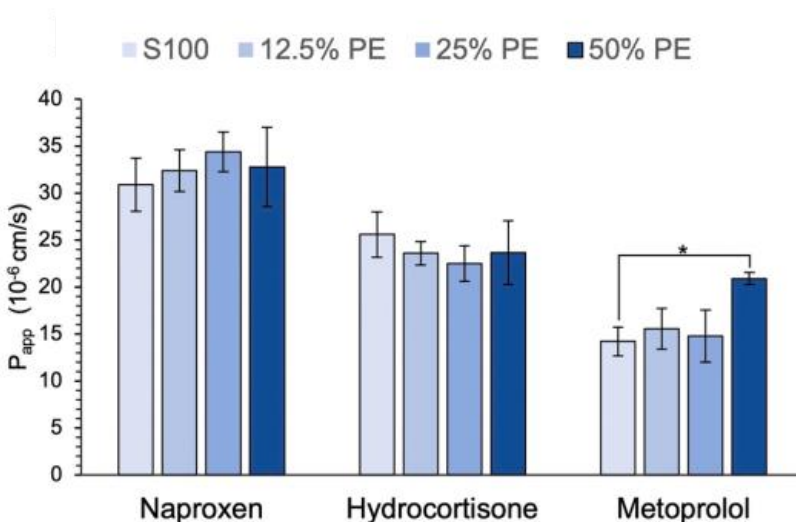


Permeapad®: Influence of phospholipid composition

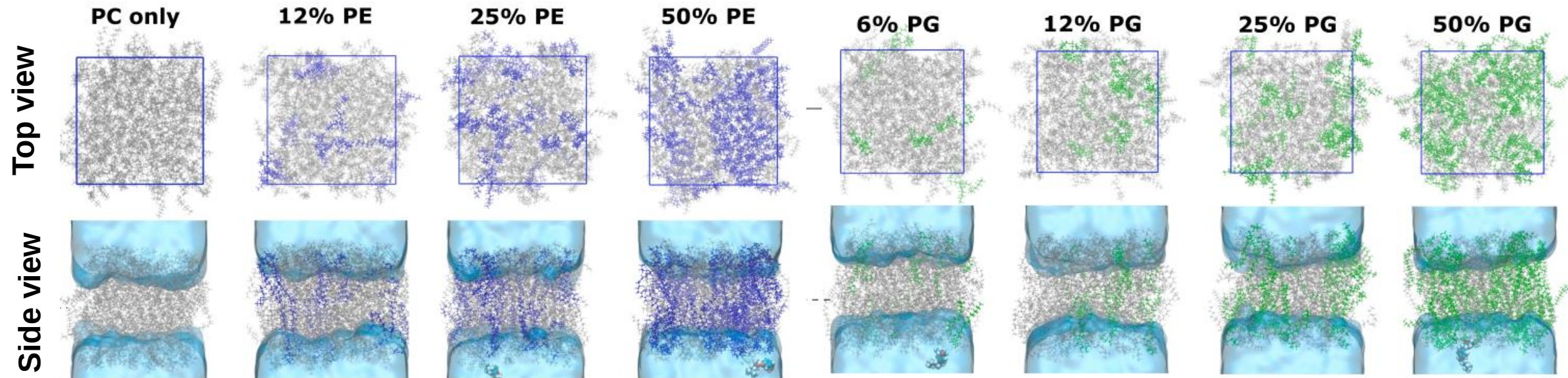
- Original and commercial Permeapad consist of 100% PC from soy (Lipoid S100)
- Series of Permeapad with 12.5-50 % phosphatidylethanolamine, 6.25-50 % phosphatidylglycerol, and 50-93.75 % phosphatidylcholine
- 100% Lipoid E80

Significant effects:

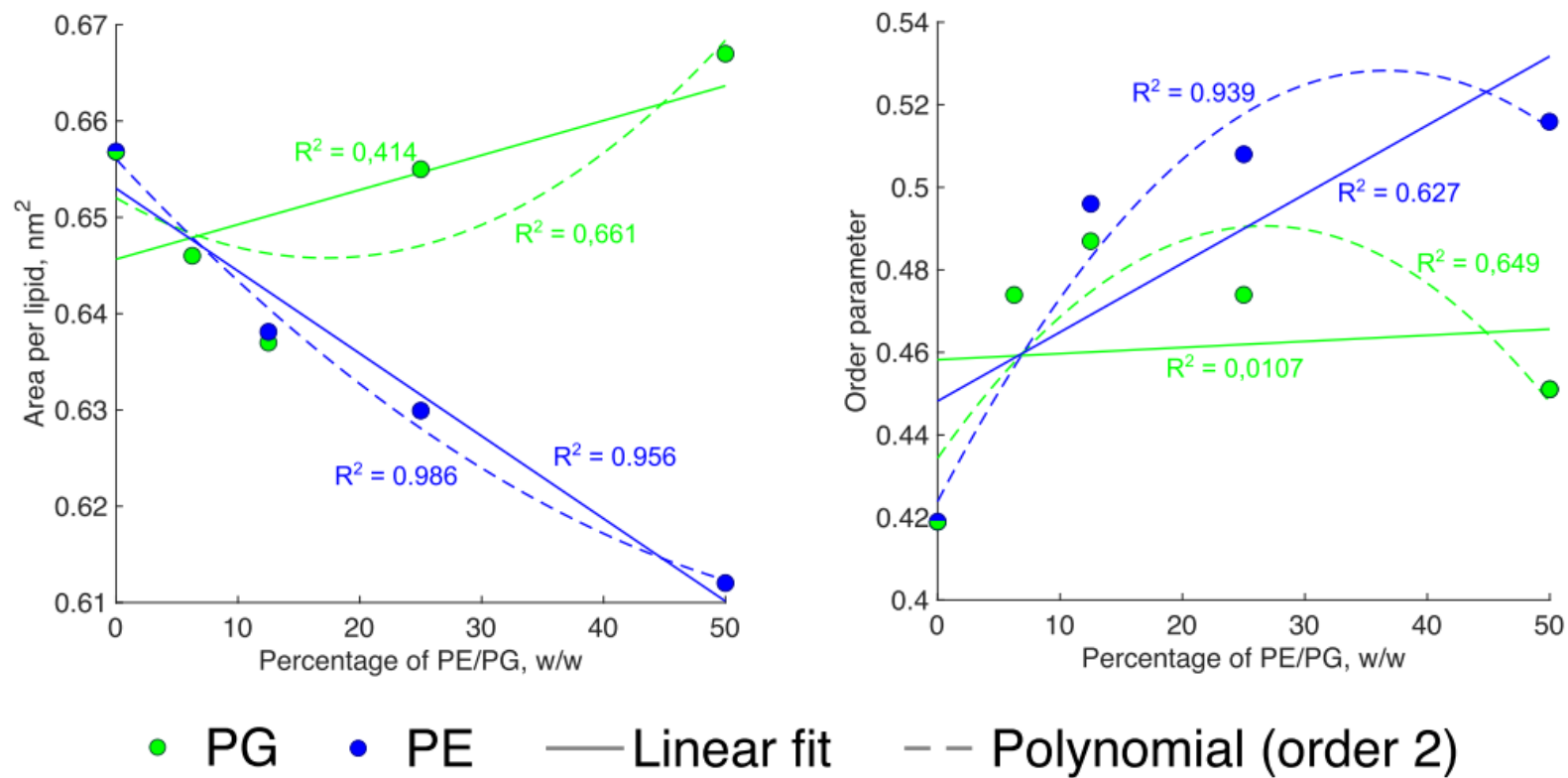
- 50 % PE, metoprolol
- 6.25 % PG, metoprolol and hydrocortisone
- 12.5 % PG, metoprolol



Molecular dynamics simulations: PE, PG membranes

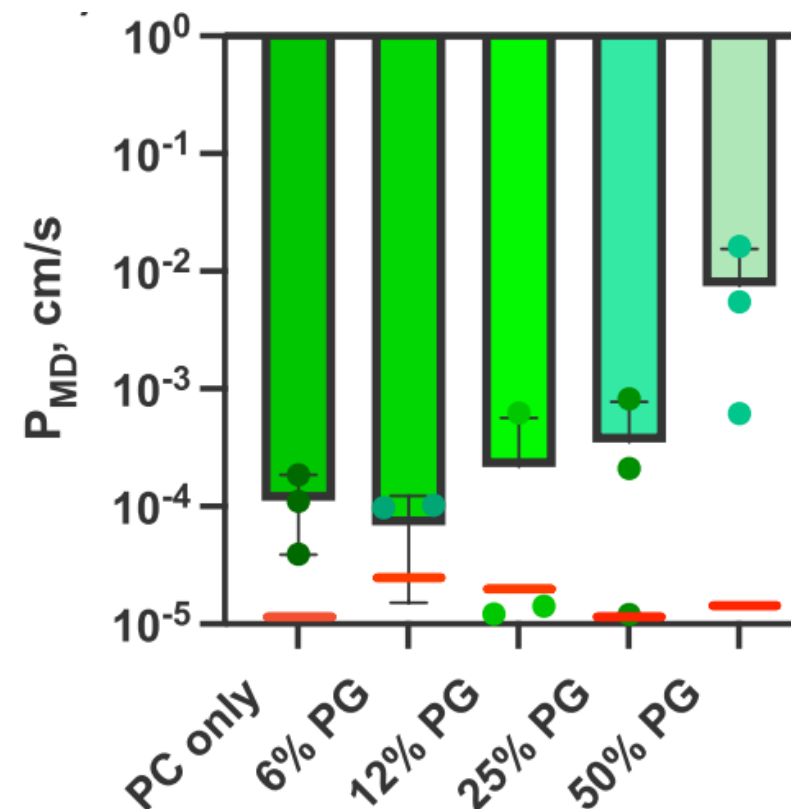
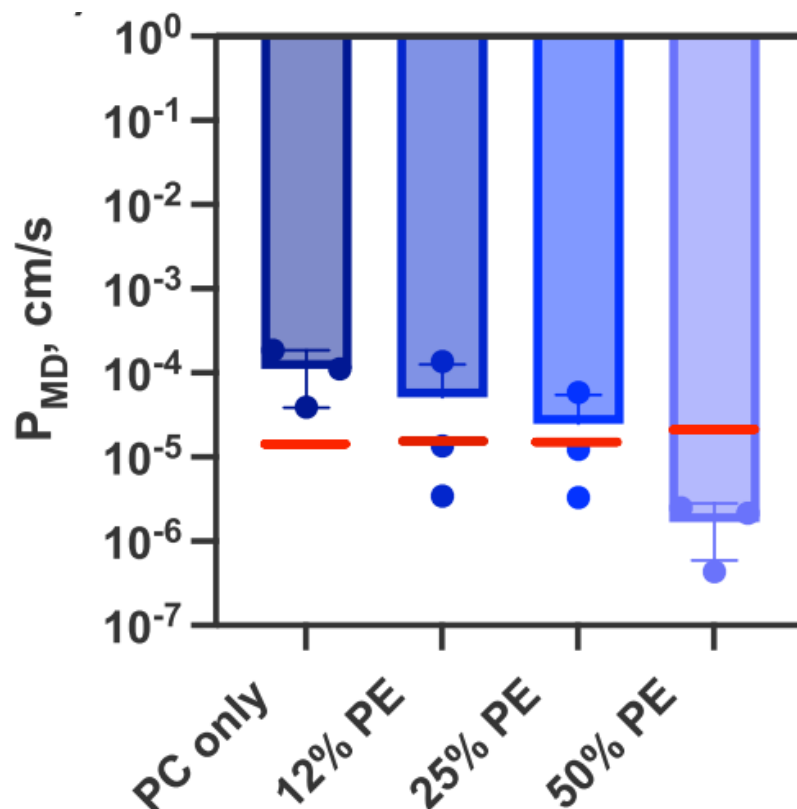


Molecular dynamics simulations: PE, PG membranes

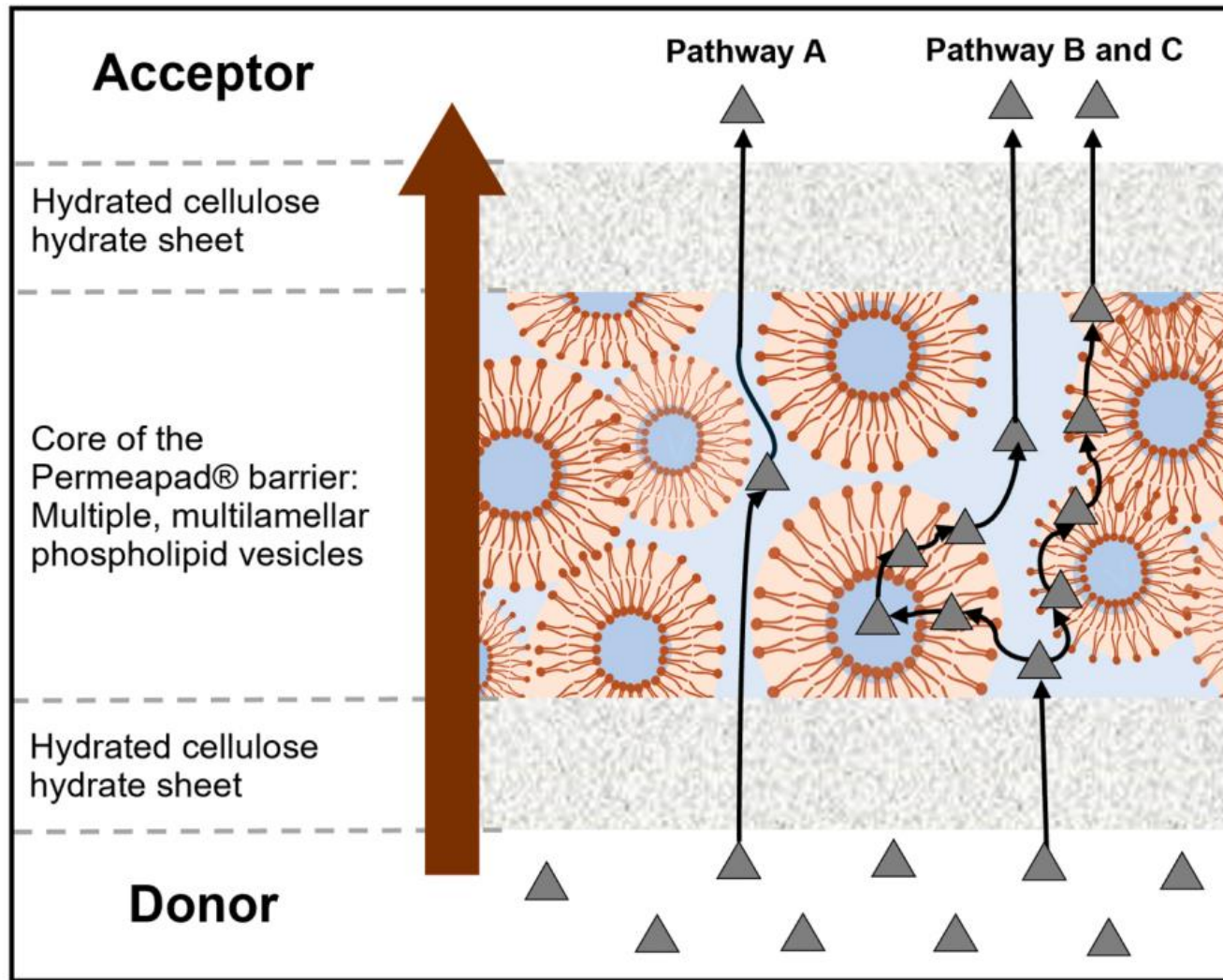


Molecular dynamics simulations: Metoprolol and PE, PG membranes

- In silico permeability values calculated from free energy profiles and diffusivity values at various distances from the membrane (umbrella sampling simulations)



Permeation pathways in Permeapad

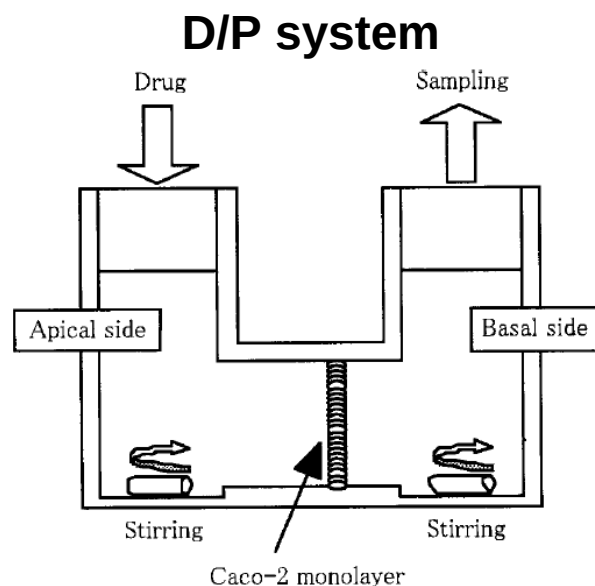


- Permeapad® is more complex than a single phospholipid bilayer
- Type of phospholipid in Permeapad® can affect:
 1. How drugs are attracted to the liposome surface,
 2. whether they will get internalized into the liposomes,
 3. whether they easily can redistribute.

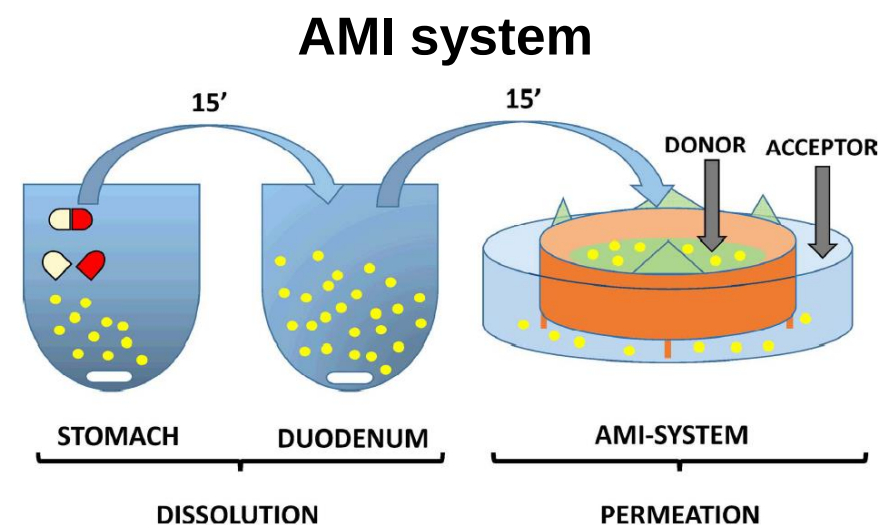
Phospholipids in combined dissolution/permeation studies

Combined dissolution/permeation studies

- Dissolution in an absorptive environment
- Performance evaluation of enabling formulations (e.g. amorphous solid dispersions; ASDs)
- **Basic set-up:**
Two vessels, a donor and an acceptor, separated by a permeation barrier
- Simultaneous or sequential



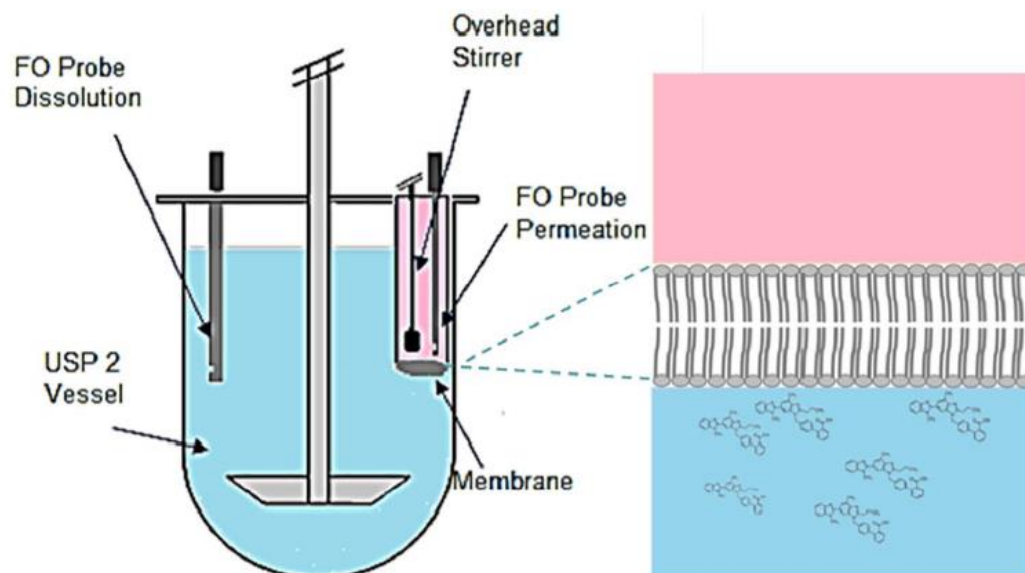
Kataoka, M., et al., 2003. Pharm. Res. 20, 1674–1680.



Berben, P., et al. 2018, Int. J. Pharm. 537, 22–29.

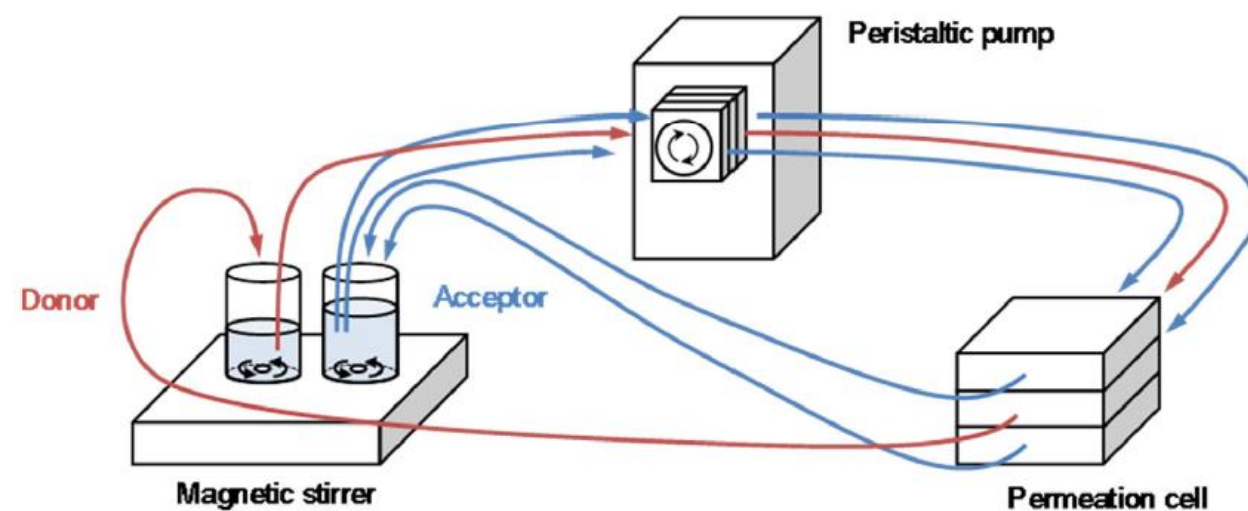
Combined dissolution/permeation studies

macroFLUX™



Borbás, E., et al., 2018. Eur. J. Pharm. Sci. 114, 310–317.

Permealoo™



Sironi, D., et al., 2018. J. Pharm. Biomed. Anal. 156, 247–251.

- Overview of set-ups in O'shea et al. 2022, DOI: [10.1016/j.ejps.2021.106098](https://doi.org/10.1016/j.ejps.2021.106098)

Example:

Influence of phospholipids on performance evaluation of ASDs

Dissolution of ASD

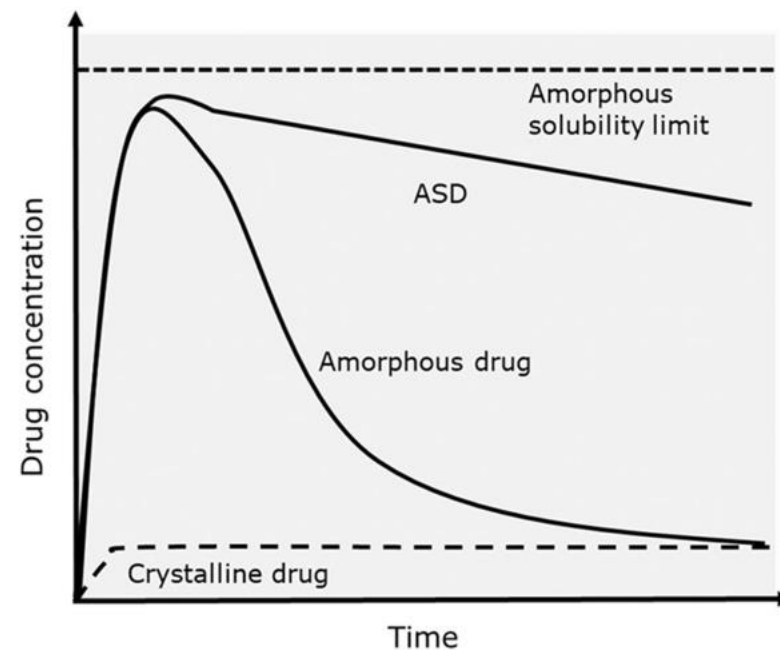
Transient
supersaturation

Increased driving
force for absorption



Effect of phospholipids on supersaturation?

- Nucleation induction time in presence/absence of phospholipids (a type of solvent shift assay)
- Combined dissolution/permeation studies in presence/absence of phospholipids



Effect of phospholipids on nucleation induction time (NIT)

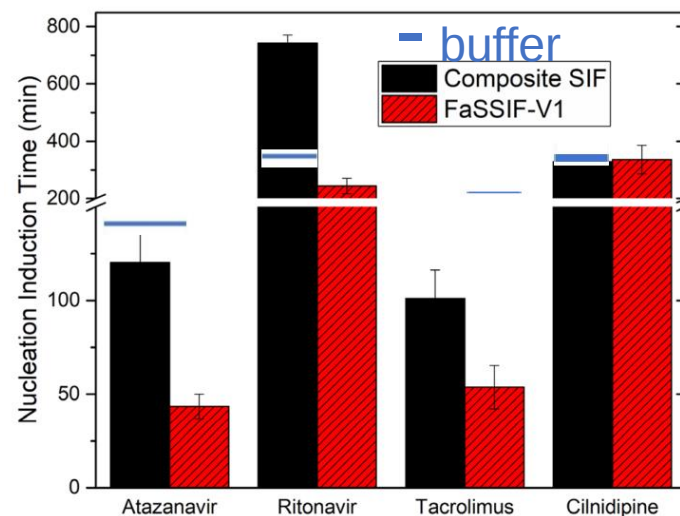
Drug in
methanol

Aliquot in test medium
(e.g., FaSSIF V1)

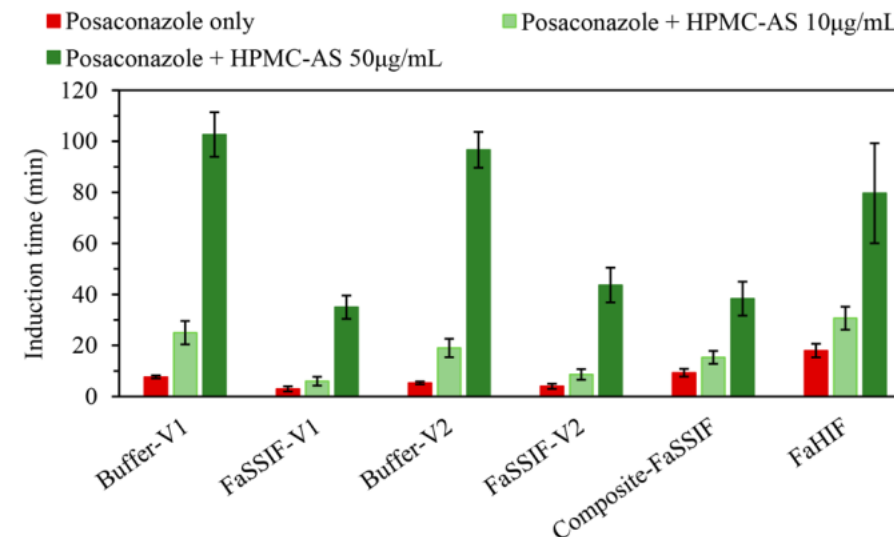
UV/vis spectrophotometer
with fiber optic dip probe:
measure extinction at non-
absorbing wavelength

Increase in
scattering: NIT

- Presence of mixed micelles with lecithin (e.g. FaSSIF V1) decreases NIT of some drugs (but not all)
- Lyso-PC more potent than PC (atazanavir)
- Effect dependent on (atazanavir):
 - Concentration
 - Presence of bile salts

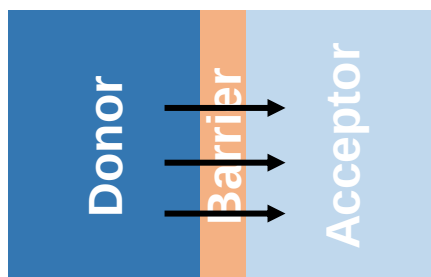


Indulkar et al. 2018, *Pharm. Res.*,
DOI: 10.1007/s11095-018-2441-2



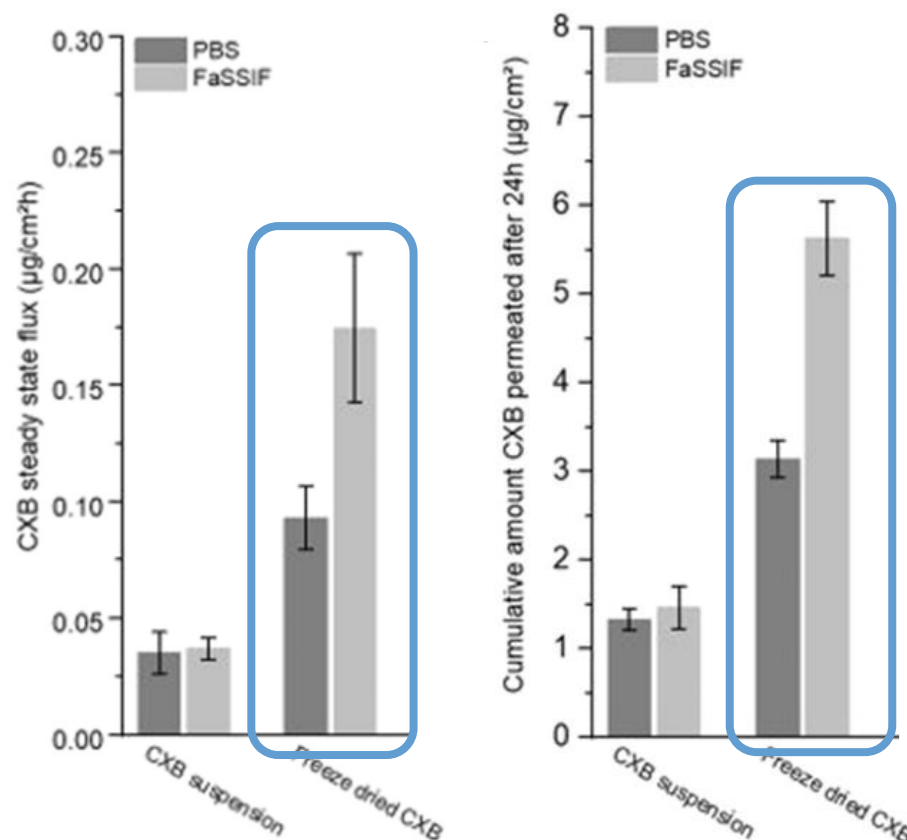
Elkhabaz et al. 2021, *Cryst. Growth Des.*,
DOI: 10.1021/acs.cgd.0c01730

Effect of phospholipids on permeation from amorphous solids

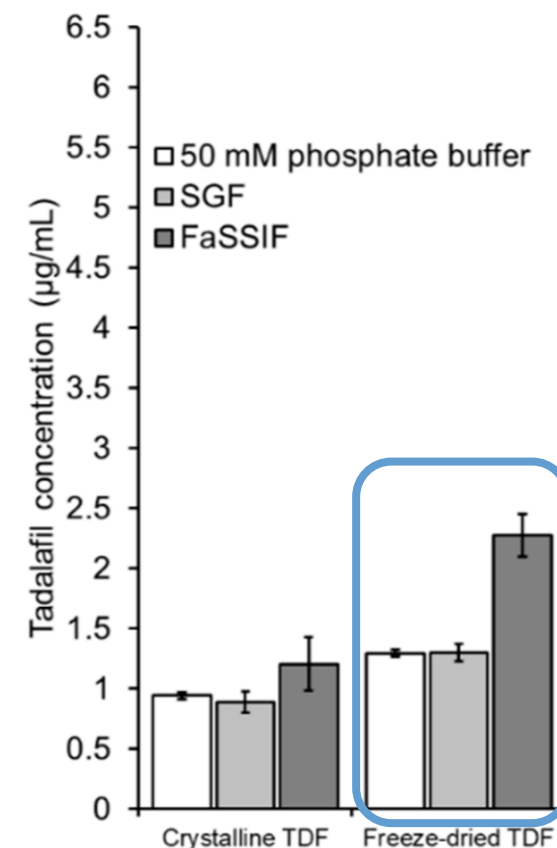


- Permeation as indirect measure for supersaturation
- Longer and/or more pronounced supersaturation
→ higher permeation
- Presence of FaSSiF V1 increased permeation from amorphous celecoxib and tadalafil

Celecoxib



Tadalafil



Thank you for your attention!

Acknowledgements

- Asta Kabels
- Mikkel Højmark Tønning
- Sune Nielsen
- Helena Holm Trabjerg
- Dr. Aleksei Kabedev
- Dr. Felix Paulus
- Dr. Philipp A. Elvang
- Prof. Anna Krupa
- Prof. Alexandra Teleki
- Prof. Annette Bauer-Brandl
- Prof. Martin Brandl

Ann-Christin Jacobsen

Kiel University
Department of Pharmaceutics and Biopharmaceutics
Grasweg 9a
24118 Kiel, Germany
e-mail: ajacobsen@pharmazie.uni-kiel.de

